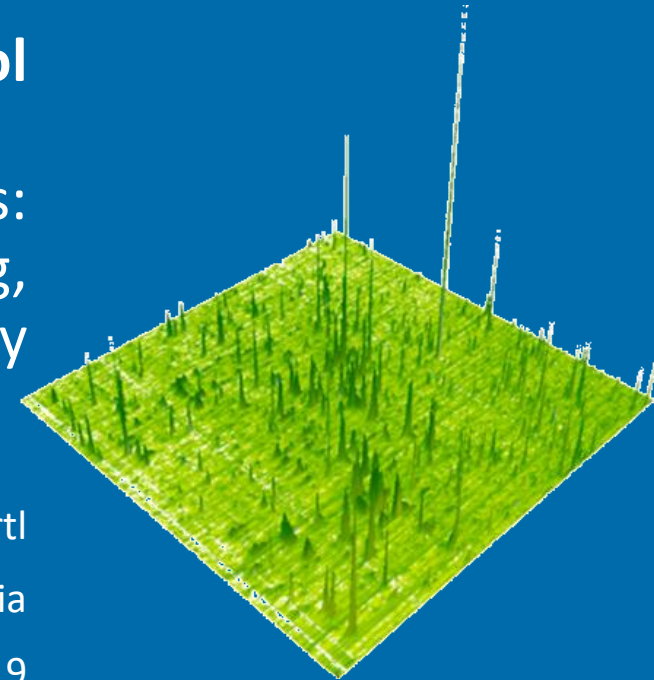


NPG-net Winter School

A primer on structural proteomics:
MS-basics, cross-linking,
HDX, ion-mobility



Markus Hartl

MFPL Mass Spectrometry Facility, Vienna, Austria

Talk @ CEITEC, Brno, January 10, 2019

The purpose of this lecture

Explore the basics of mass spectrometry and get an idea of what it can do for you in terms of:

- Proteoform identification
- Structural information
- Protein dynamics

...and be honest about what the difficulties are.

What mass spectrometry is all about.

Analytical balances:

0.001 g to 1 g \pm 0.0001 g



Mass spectrometers:

1E-24 g to 1E-19 g \pm 1E-26 g

or

1 Da to 100.000 Da \pm 0.01 Da

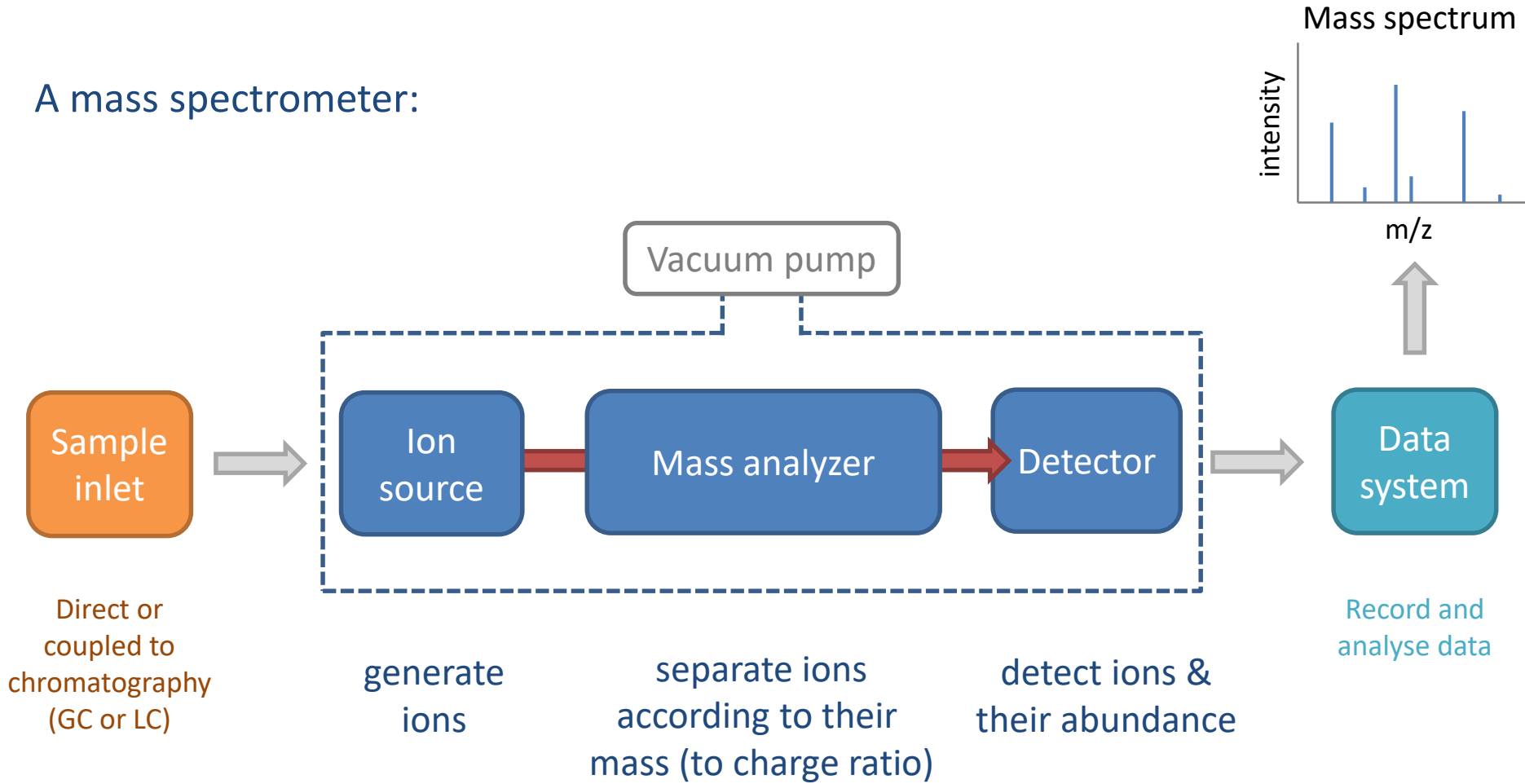
Why is that interesting?

- Identify or verify compounds by their discrete atomic mass.
- Quantify compounds
- Gain information on structure/sequence

- Target analytes: biomolecules (metabolites, oligonucleotides, peptides, proteins), synthetic chemicals, polymers, drugs, etc.

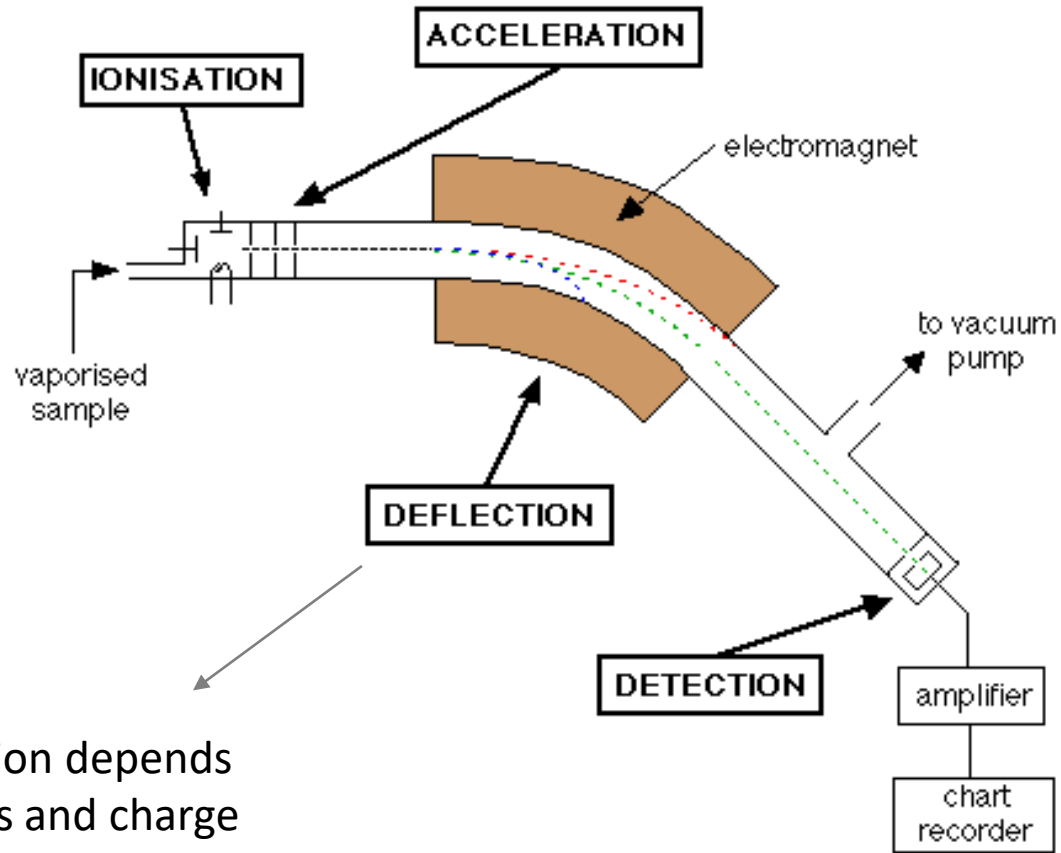
How does it work?

A mass spectrometer:

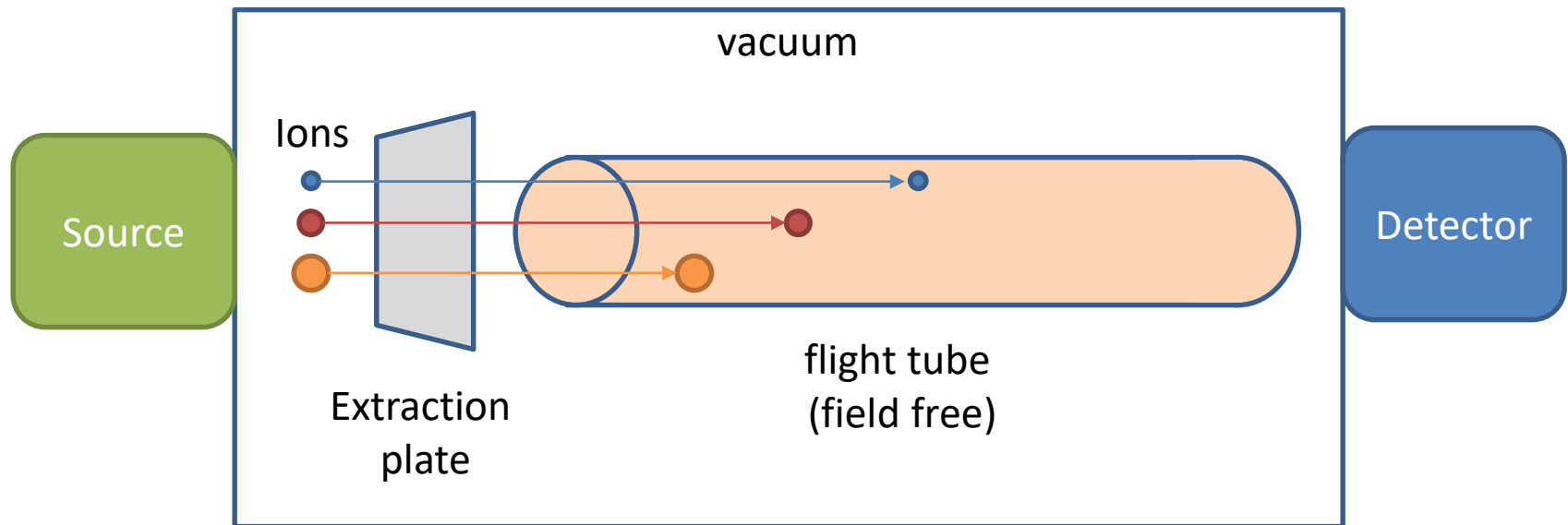


How does it work?

(a sector-field instrument as an example)

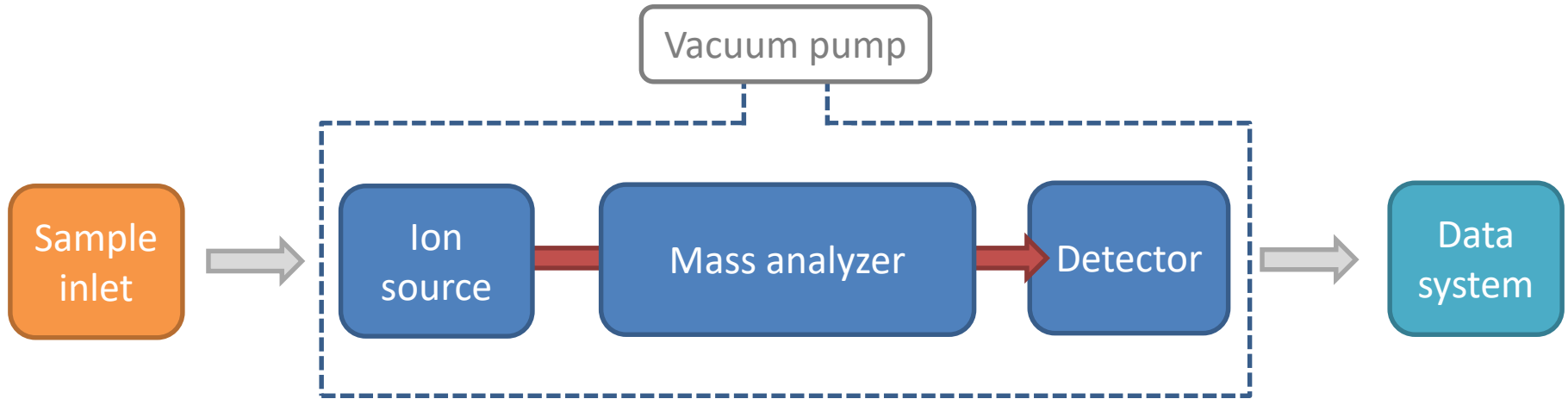


Time-of-flight analyzer



$$E_p = q \cdot U \quad \rightarrow \quad E_k = \frac{m \cdot v^2}{2} \quad \leftrightarrow \quad v = \sqrt{\frac{2 \cdot Ek}{m}} \quad \leftrightarrow \quad \frac{d}{t} = \sqrt{\frac{2 \cdot Ek}{m}} \quad \rightarrow \quad \frac{m}{q} = \frac{2 \cdot U \cdot t^2}{d^2}$$

How does it work?



generate ions

separate ions according to their mass (to charge ratio)

detect ions & their abundance

For example:

Electron impact (EI)

Chemical ionis. (CI)

Electrospray ionis. (ESI)

MALDI

} **hard**

} **soft**

For example:

Quadrupol

Iontrap

Time-of-Flight (TOF)

Orbitrap

} **Low res**


} **High res**

Instrument dependent:

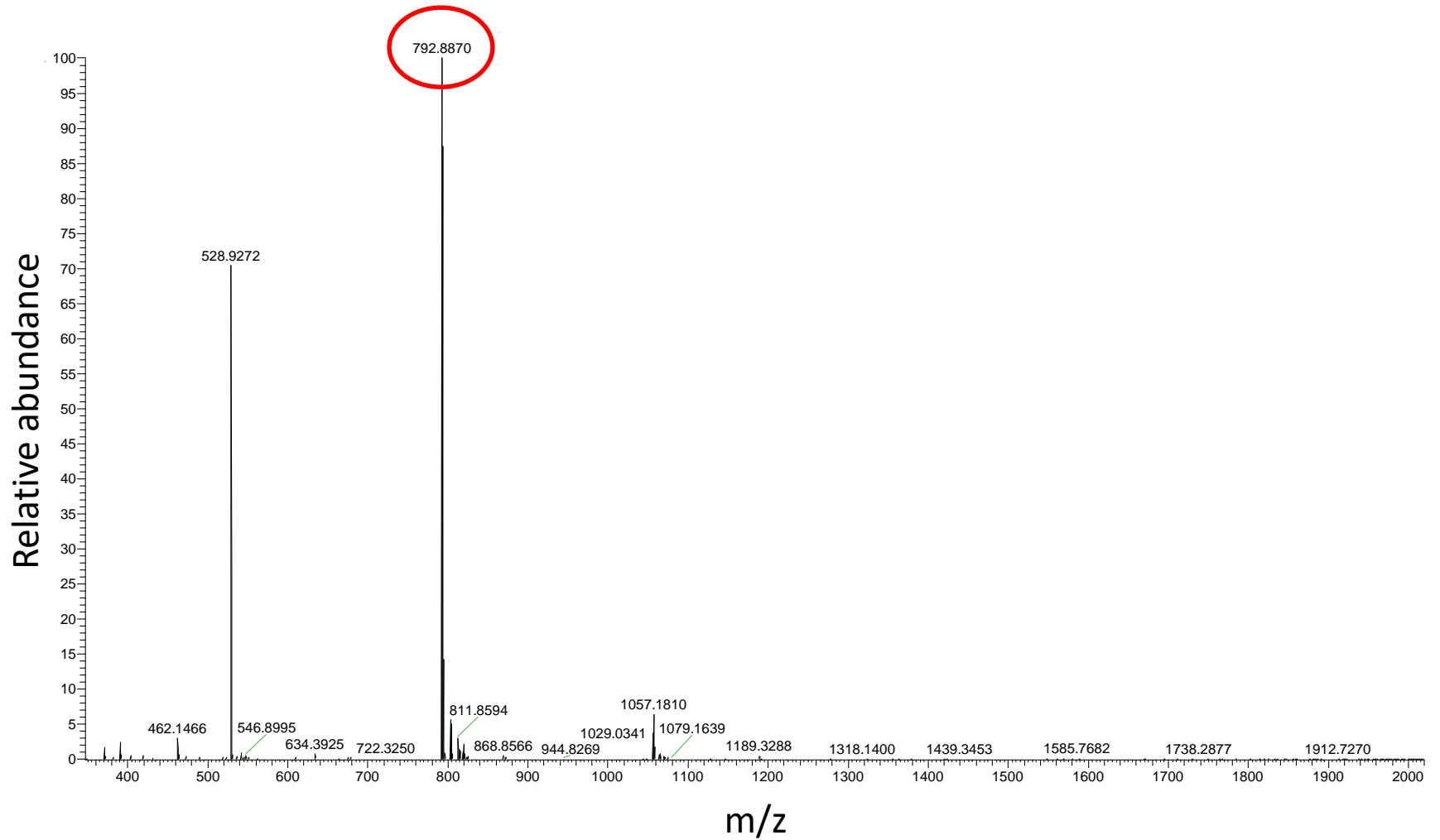
e.g. Electron multiplier

Many combinations many acronyms

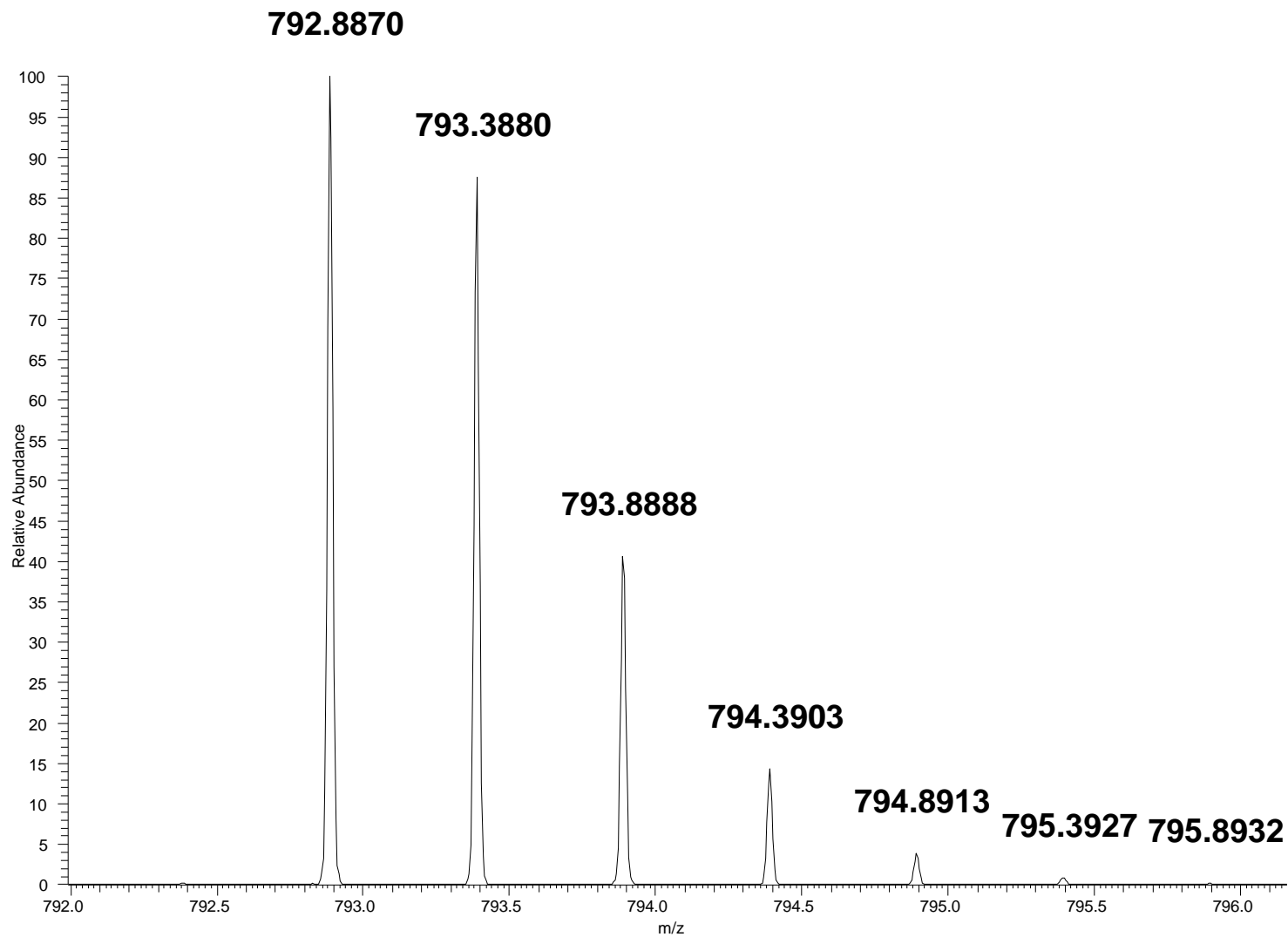
Separation	Ionisation	Mass Spectrometer type	
LC	ESI	TOF	LC-ESI-TOF
GC	(EI)	Quadrupole (Q)	GC-Quad
-	MALDI	TOF	MALDI-TOF
LC	(ESI)	Triple-Quadrupole (QQQ)	LC-TripleQ
LC	(ESI)	Quadrupole-Orbitrap	LC-Q Exactive
....			

 Different instrumental setups for different questions.

The output: mass spectra



Zoom on signal at 792.8870 m/z



Atomic Weights and Isotopic Compositions for All Elements

<u>Isotope</u>		<u>Relative Atomic Mass</u>	<u>Isotopic Composition</u>	
1	H	1	1.007 825 032 07(10)	0.999 885(70)
	D	2	2.014 101 777 8(4)	0.000 115(70)
	T	3	3.016 049 2777(25)	
6	C	12	12.000 000 0(0)	0.9893(8)
		13	13.003 354 8378(10)	0.0107(8)
		14	14.003 241 989(4)	
7	N	14	14.003 074 004 8(6)	0.996 36(20)
		15	15.000 108 898 2(7)	0.003 64(20)
8	O	16	15.994 914 619 56(16)	0.997 57(16)
		17	16.999 131 70(12)	0.000 38(1)
		18	17.999 161 0(7)	0.002 05(14)
15	P	31	30.973 761 998 42(70)	1
16	S	32	31.972 071 1744(14)	0.9499(26)
		33	32.971 458 9098(15)	0.0075(2)
		34	33.967 867 004(47)	0.0425(24)
		36	35.967 080 71(20)	0.0001(1)

Defining mass

Defining mass (according to IUPAC, <https://goldbook.iupac.org/>):

Unified atomic mass unit:

- $\frac{1}{12}$ of the mass of a carbon-12 atom in its ground state
- $1 \text{ u} \approx 1.660\,5402\,10 \times 10^{-27} \text{ kg}$
- Symbols: u or Da (for its equivalent dalton).

Relative atomic mass: mass of a discrete atomic particle or molecule expressed in unified atomic mass units.

Standard atomic weight (or molecular weight or average mass): the weighted average of the masses of the naturally occurring isotopes. For example one carbon atom:

$$\frac{(98.8\% * 12.0 + 1.1\% * 13.003355)}{100\%} = 12.011$$

Standard atomic weight

hydrogen 1 H 1.0079																	helium 2 He 4.0026																
lithium 3 Li 6.941	beryllium 4 Be 9.0122																	boron 5 B 10.811	carbon 6 C 12.011	nitrogen 7 N 14.007	oxygen 8 O 15.999	fluorine 9 F 18.998	neon 10 Ne 20.180										
sodium 11 Na 22.990	magnesium 12 Mg 24.305																	aluminium 13 Al 26.982	silicon 14 Si 28.086	phosphorus 15 P 30.974	sulfur 16 S 32.065	chlorine 17 Cl 35.453	argon 18 Ar 39.948										
potassium 19 K 39.098	calcium 20 Ca 40.078	scandium 21 Sc 44.956	titanium 22 Ti 47.867	vanadium 23 V 50.942	chromium 24 Cr 51.996	manganese 25 Mn 54.938																	zinc 30 Zn 65.39	gallium 31 Ga 69.723	germanium 32 Ge 72.61	arsenic 33 As 74.922	selenium 34 Se 78.96	bromine 35 Br 79.904	krypton 36 Kr 83.80				
rubidium 37 Rb 85.468	strontium 38 Sr 87.62	yttrium 39 Y 88.906	zirconium 40 Zr 91.224	niobium 41 Nb 92.906	molybdenum 42 Mo 95.94	technetium 43 Tc [98]																	cadmium 48 Cd 112.41	indium 49 In 114.82	tin 50 Sn 118.71	antimony 51 Sb 121.76	tellurium 52 Te 127.60	iodine 53 I 126.90	xenon 54 Xe 131.29				
caesium 55 Cs 132.91	barium 56 Ba 137.33	lanthanoids 57-70 *	lutetium 71 Lu 174.97	hafnium 72 Hf 178.49	tantalum 73 Ta 180.95	tungsten 74 W 183.84	rhenium 75 Re [186]																	mercury 80 Hg 200.59	thallium 81 Tl 204.38	lead 82 Pb 207.2	bismuth 83 Bi 208.98	polonium 84 Po [209]	astatine 85 At [210]	radon 86 Rn [222]			
francium 87 Fr [223]	radium 88 Ra [226]	actinoids 89-102 **	lawrencium 103 Lr [262]	rutherfordium 104 Rf [261]	dubnium 105 Db [262]	seaborgium 106 Sg [266]	bohrium 107 Bh [264]	hassium 108 Hs [269]	meitnerium 109 Mt [268]	unbinilium 110 Uun [271]	ununilium 111 Uuun [272]	ununbium 112 Uub [277]																	copernicium 112 Cn [285]	nihonium 113 Nh [284]	flerovium 114 Fle [289]	tennessine 115 Ts [289]	oganeson 116 Og [289]

*lanthanoids

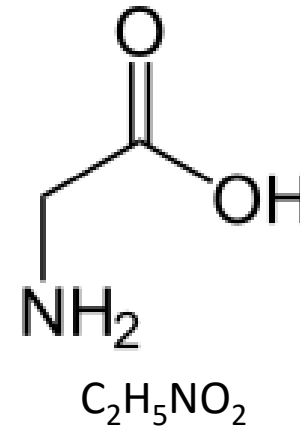
**actinoids

lanthanum 57 La 138.91	cerium 58 Ce 140.12	praseodymium 59 Pr 140.91	neodymium 60 Nd 144.24	promethium 61 Pm [145]	samarium 62 Sm 150.36	europium 63 Eu 151.96	gadolinium 64 Gd 157.25	terbium 65 Tb 158.93	dysprosium 66 Dy 162.50	holmium 67 Ho 164.93	erbium 68 Er 167.26	thulium 69 Tm 168.93	ytterbium 70 Yb 173.04
actinium 89 Ac [227]	thorium 90 Th 232.04	protactinium 91 Pa 231.04	uranium 92 U 238.03	neptunium 93 Np [237]	plutonium 94 Pu [244]	americium 95 Am [243]	curium 96 Cm [247]	berkelium 97 Bk [247]	californium 98 Cf [251]	einsteinium 99 Es [252]	fermium 100 Fm [257]	mendelevium 101 Md [258]	nobelium 102 No [259]

Example 1:

Atomic Weights and Isotopic Compositions for All Elements

Isotope	Relative Atomic Mass	Isotopic Composition	Standard Atomic Weight	Notes
1 H	1.007 825 032 07(10)	0.999 885(70)	[1.007 84, 1.008 11]	m
D	2.014 101 777 8(4)	0.000 115(70)		
T	3.016 049 2777(25)			
2 He	3.016 029 3191(26)	0.000 001 34(3)	4.002 602(2)	g,r
	4.002 603 254 15(6)	0.999 998 66(3)		
3 Li	6.015 122 795(16)	0.0759(4)	[6.938, 6.997]	m
	7.016 004 55(8)	0.9241(4)		
4 Be	9.012 182 2(4)	1.0000	9.012 1831(5)	
5 B	10.012 937 0(4)	0.199(7)	[10.806, 10.821]	m
	11.009 305 4(4)	0.801(7)		
6 C	12.000 000 0(0)	0.9893(8)	[12.0096, 12.0116]	
	13.003 354 8378(10)	0.0107(8)		
	14.003 241 989(4)			
7 N	14.003 074 004 8(6)	0.996 36(20)	[14.006 43, 14.007 28]	
	15.000 108 898 2(7)	0.003 64(20)		
8 O	15.994 914 619 56(16)	0.997 57(16)	[15.999 03, 15.999 77]	
	16.999 131 70(12)	0.000 38(1)		
	17.999 161 0(7)	0.002 05(14)		



Glycine:

Monoisotopic: 75.03203 u

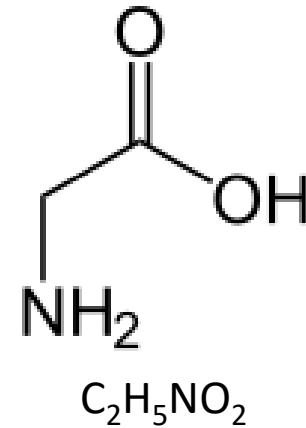
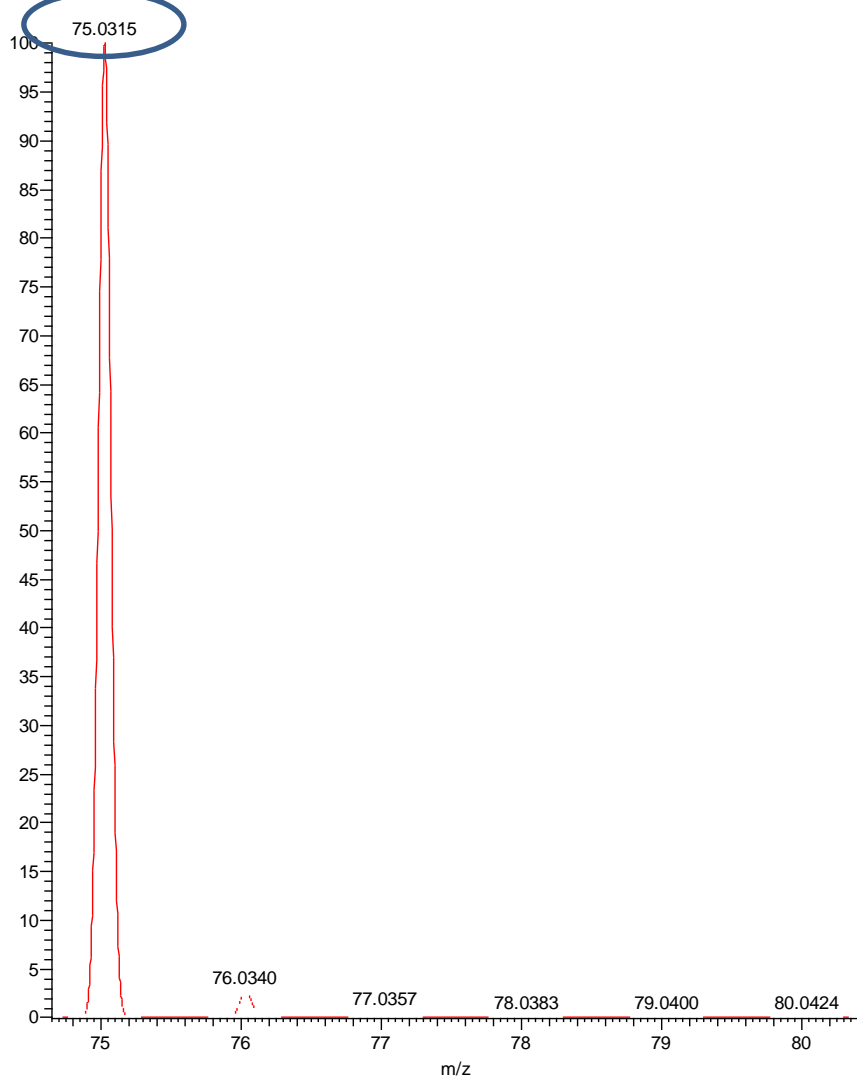
Atomic weight: 75.06720 u

Nominal mass: 75 u

Example 1:

Simulated spectrum

C₂H₅NO₂: C₂H₅N₁O₂ p(gss, s/p:40) Chrg 1R: 0.1 D...



Monoisotopic mass: 75.0315 u

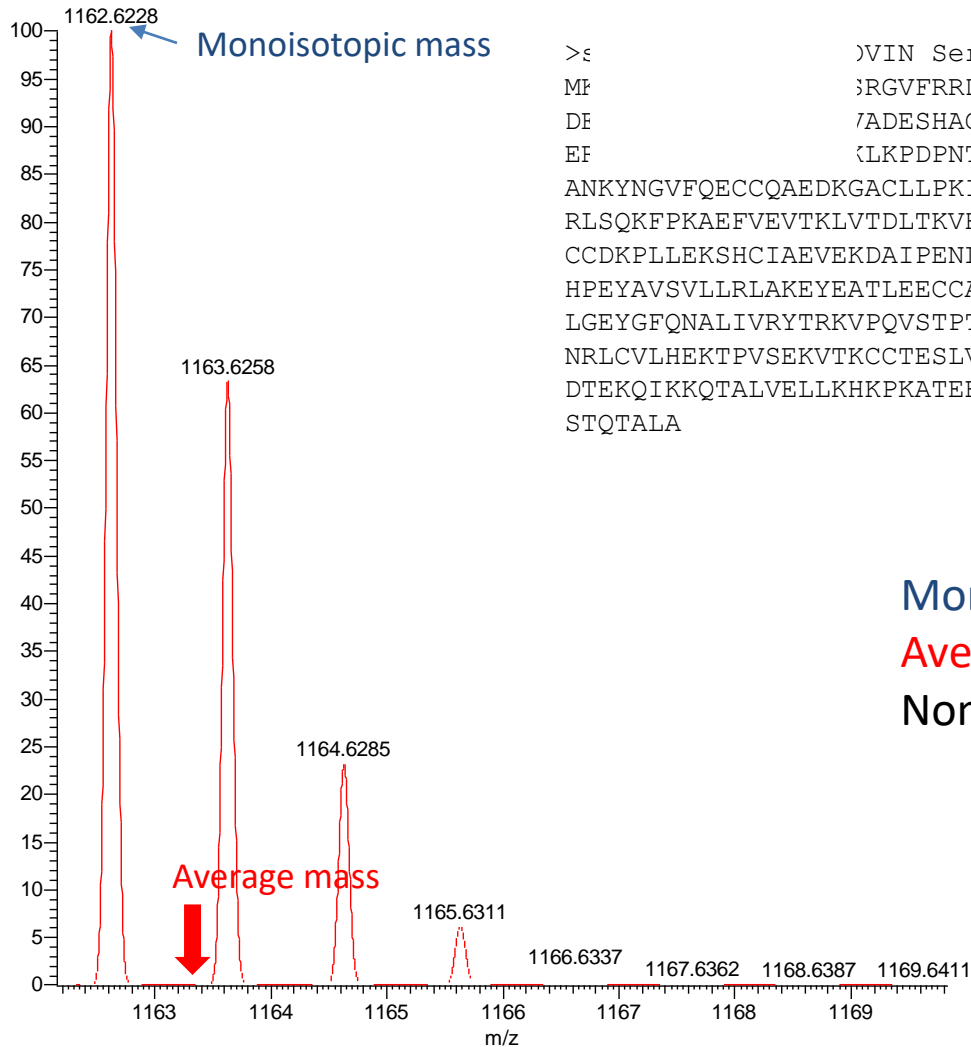
Average mass: 75.0672 u

Nominal mass: 75 u

Example 2: a peptide

Simulated spectrum

LVNELTEFAK +H2O: C53 H86 N12 O17 p(gss, s/p:40) Chrg...



Monoisotopic mass

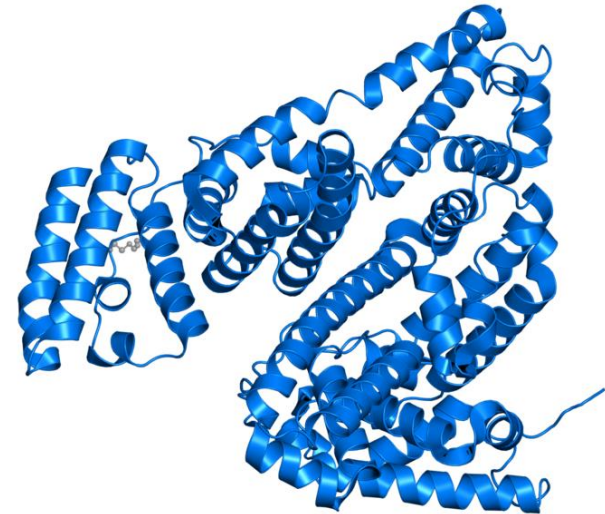
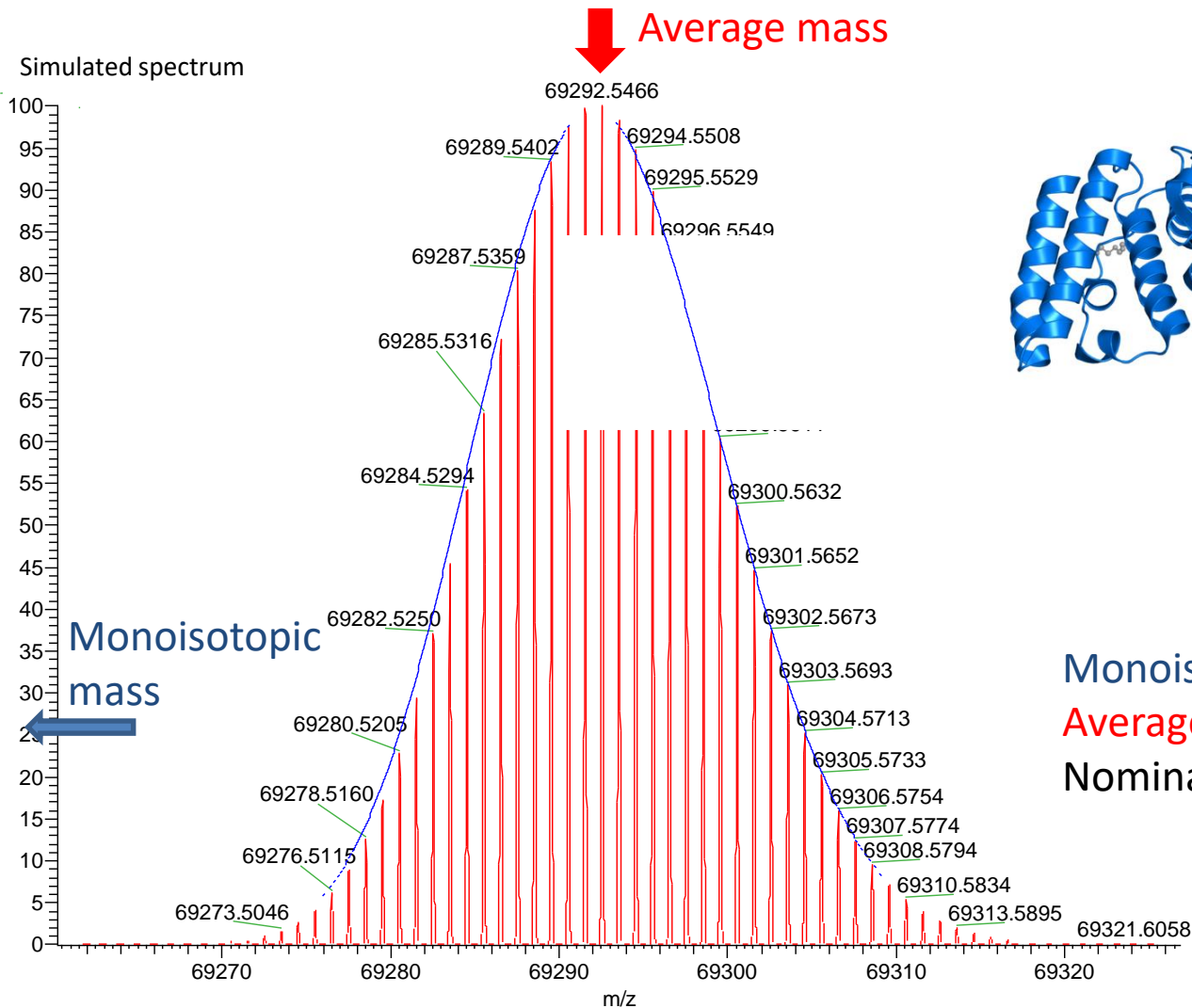
```
>ε          )VIN Serum albumin
MF          )RGVFRRDTHKSEIAHRFKDLGEEHFKGLVLIAFSQYLQQCPF
DE          )ADESHAGCEKSLHTLFGDELCKVASLRETYGDMADCCEKQEP
EF          )LKPDPNTLCDEFKKADEKKFWGKLYEIAARRHPYFYAPELLYY
ANKYNGVFQECCQAEDKGACLLPKIETMREKVLASSARQRLRCASIQKFGERALKAWSVA
RLSQKFPKAEFVEVTKLVTDLTKVHKECCHGDLLECADDRADLAKYICDNQDTISSKLKE
CCDKPLLEKSHCIAEVEKDAIPENLPPLTADFAEDKDVCKNYQEAKDAFLGSFLYEYSRR
HPEYAVSVLLRLLAKEYEATLEECCAADDPHACYSTVFDKCLKHLVDEPQNLIKQNCDOFEK
LGEYGFQNALIVRYTRKVPQVSTPTLVEVSRSLGKVGTRCCTKPESERMPCTEDYLSLIL
NRLCVLHEKTPVSEKVTKCCTESLVNRRPCFSALTPDETYVPKAFDEKLFTHADICTLP
DTEKQIKKQTALVELLKHKPKATEEQKLTVMENFVAFVDKCCAADDKEACFAVEGPKLVV
STQTALA
```

Monoisotopic mass: 1162.6228 u

Average mass: 1163.34 u

Nominal mass: 1163 u

Example 3: a protein



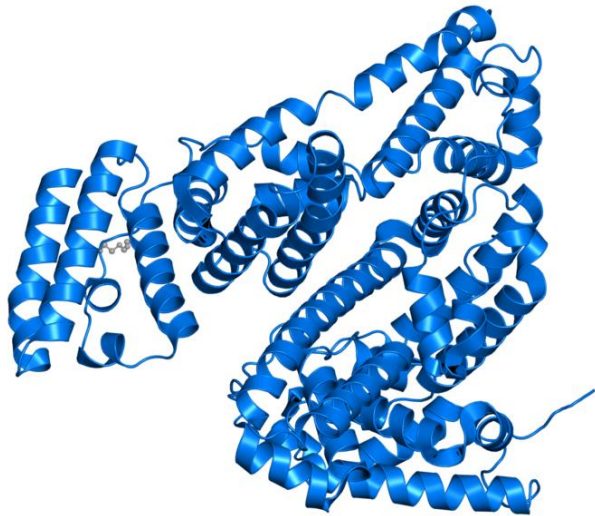
BSA

Monoisotopic mass: 69248.44 u

Average mass: 69293.41 u

Nominal mass: 69293 u

The basic idea



Bovine serum albumin
69293 Da

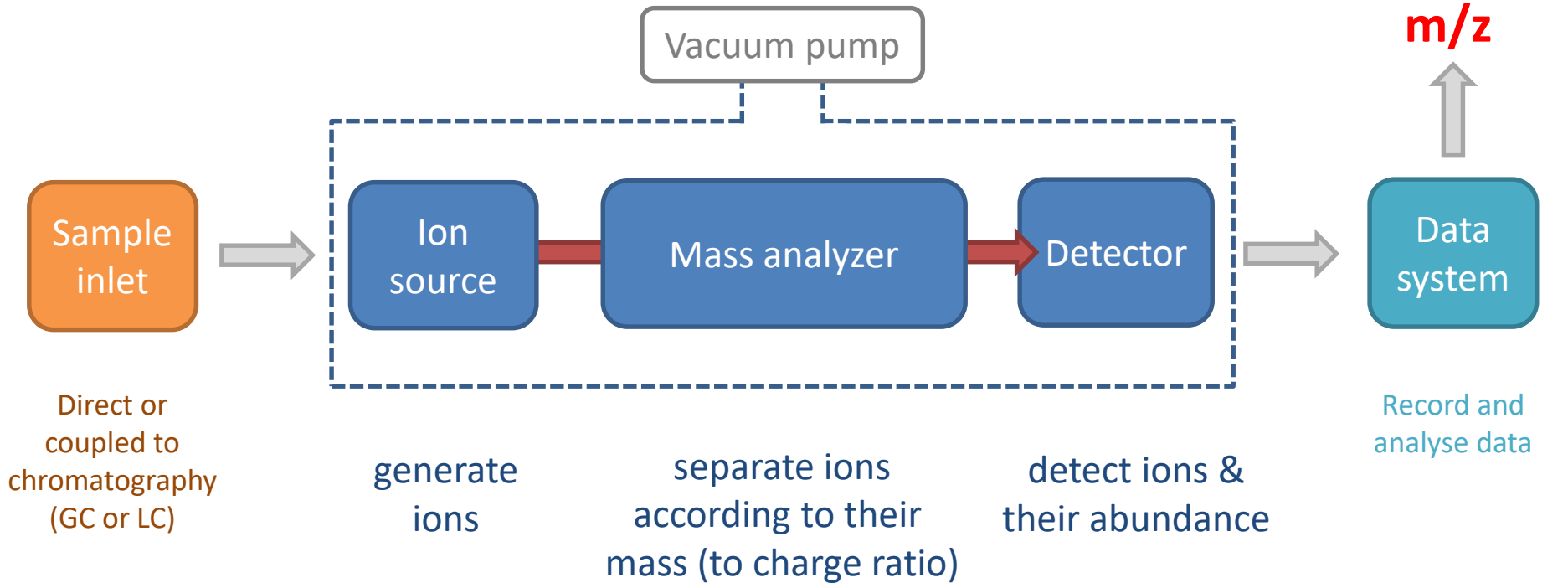


69293 Da measured

**Which mass do you measure?
Isotopic, average?
How accurately?**

MS measures mass / charge (m/z)

A mass spectrometer:



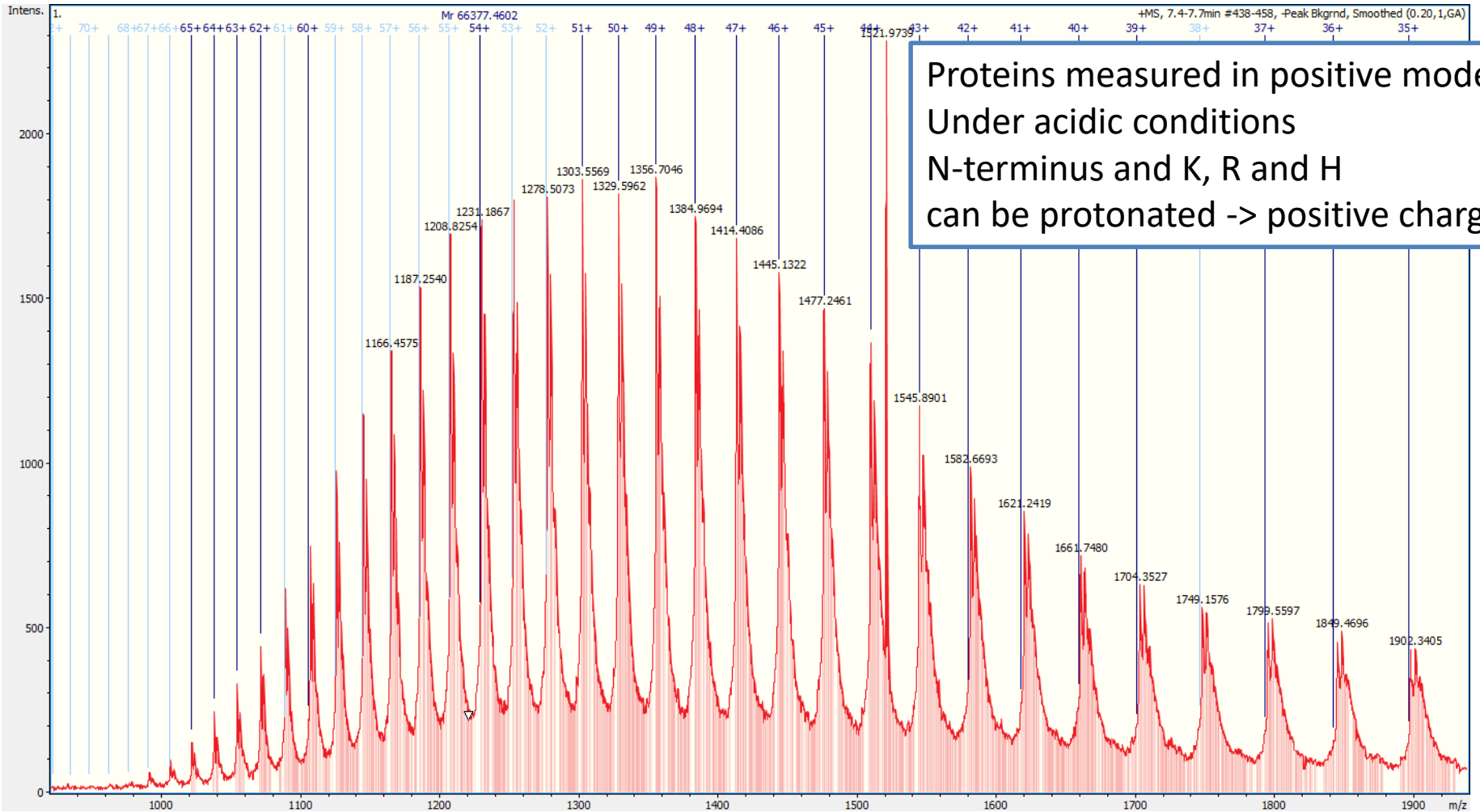
What is the charge of a protein in electrospray ionisation (ESI)?

Charge envelopes in ESI: measuring m/z

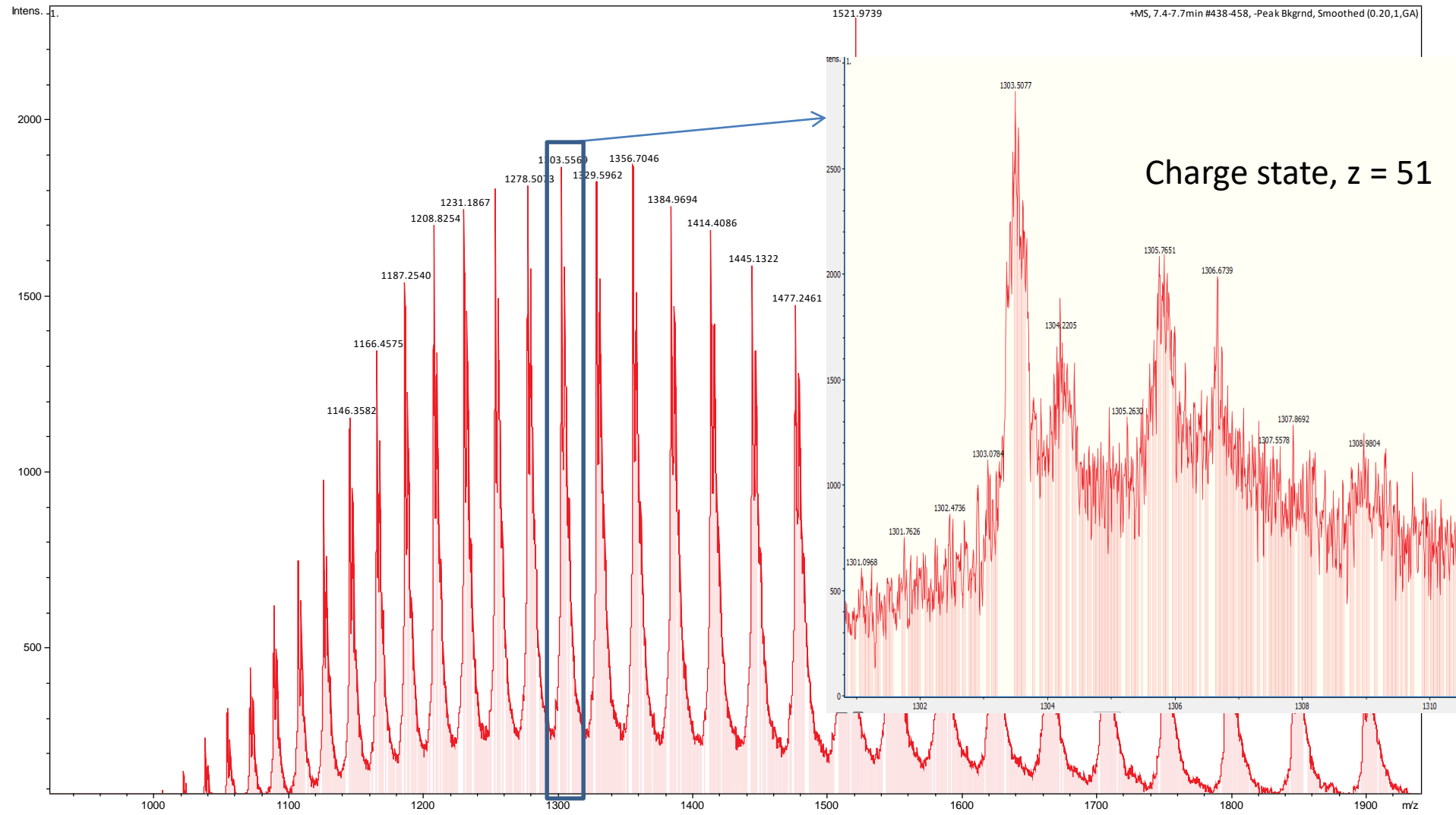
55+

50+

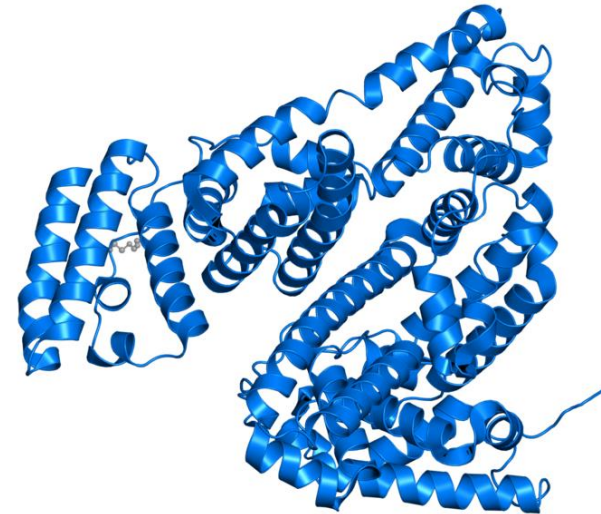
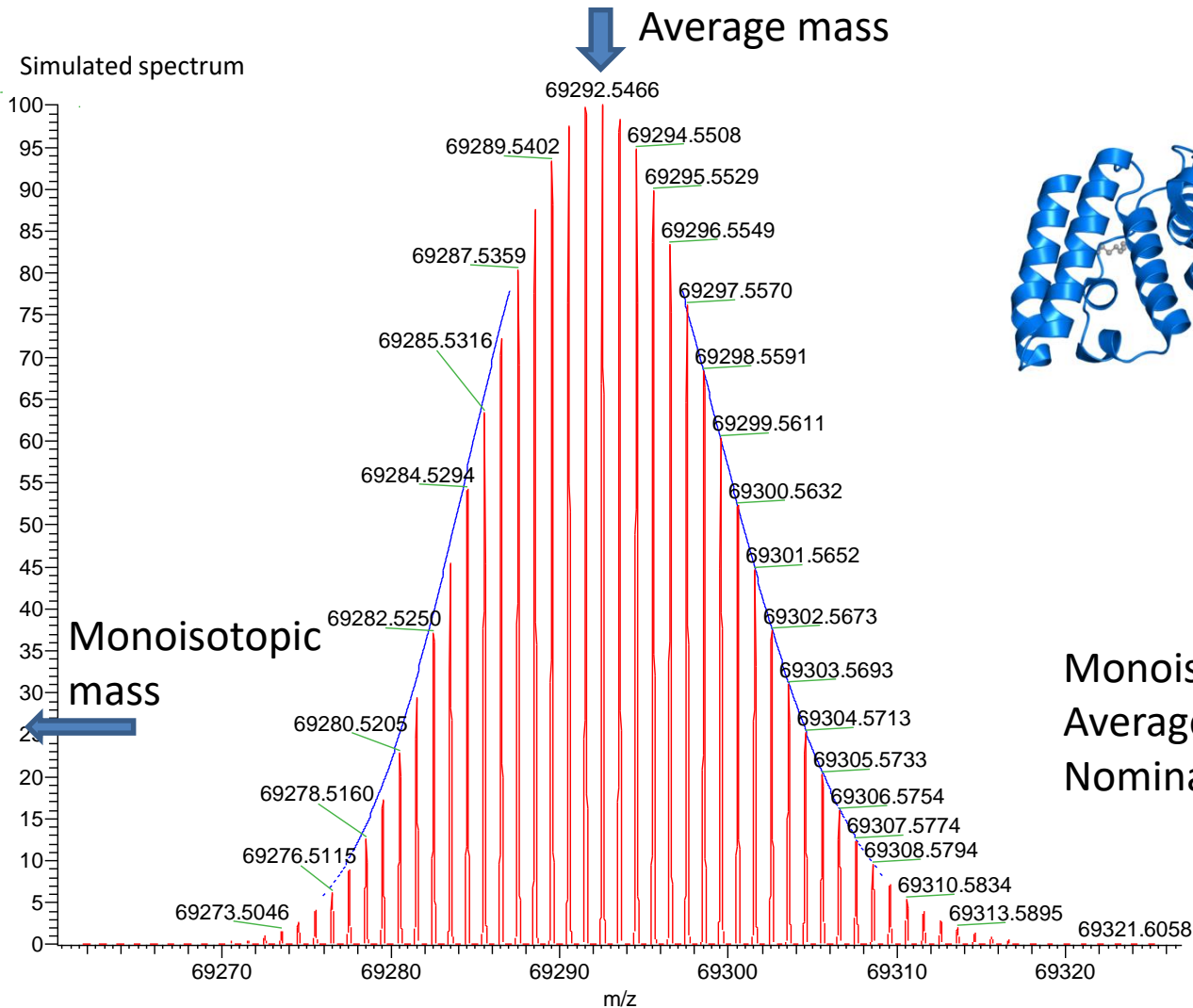
45+



ESI-TOF spectrum of intact BSA



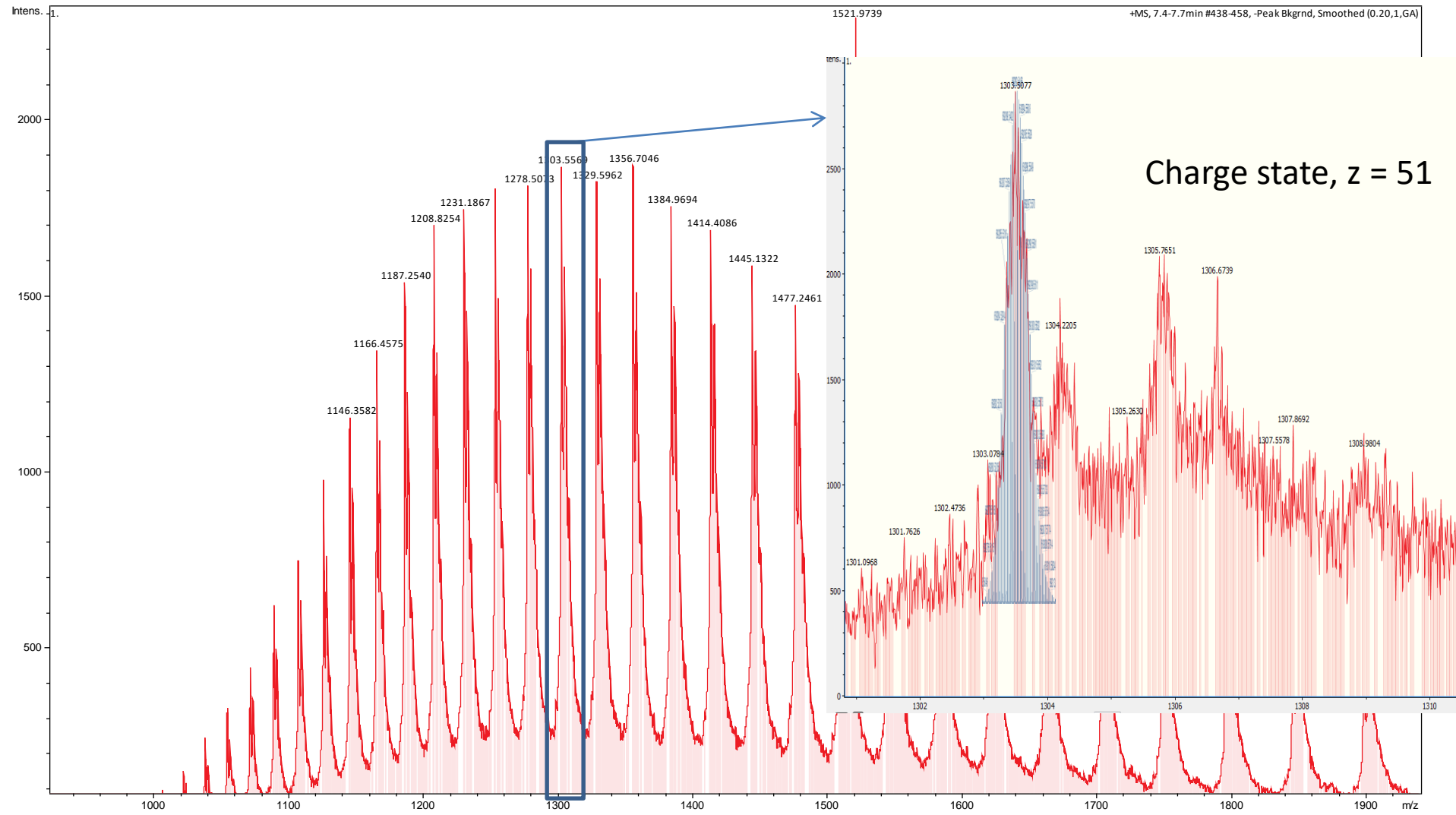
Theoretical uncharged spectrum



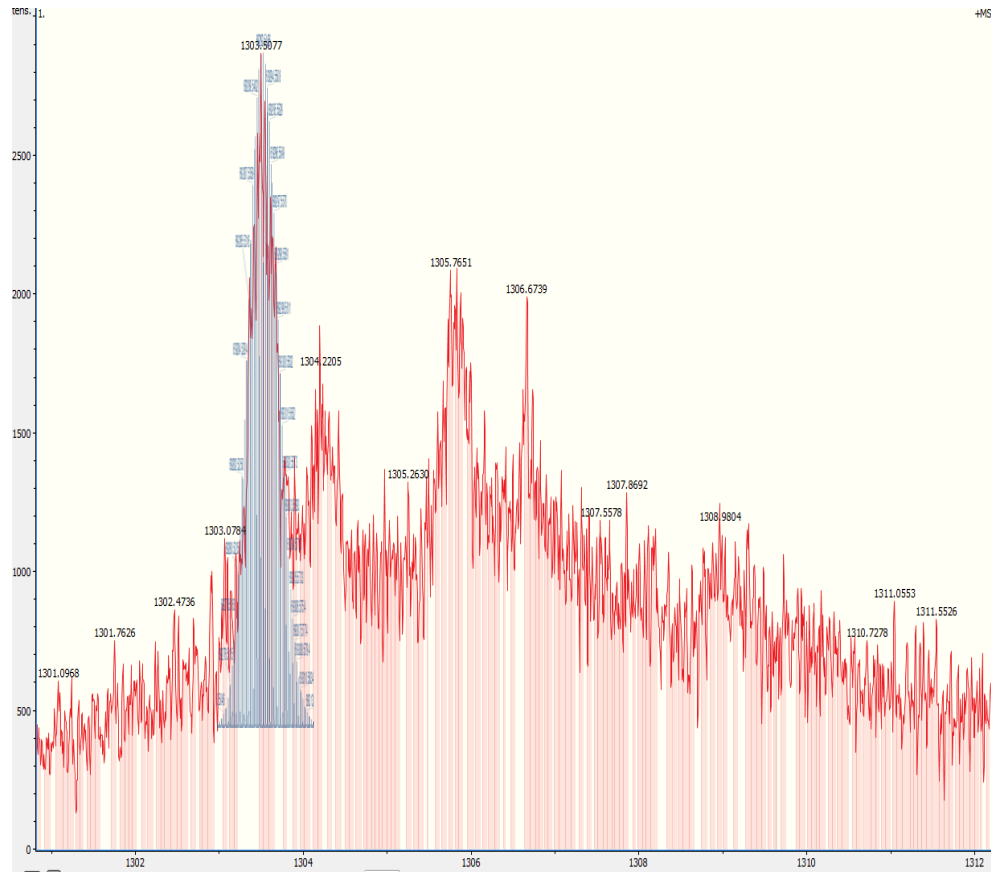
BSA

Monoisotopic mass: 69248.44 u
Average mass: 69293.41 u
Nominal mass: 69293 u

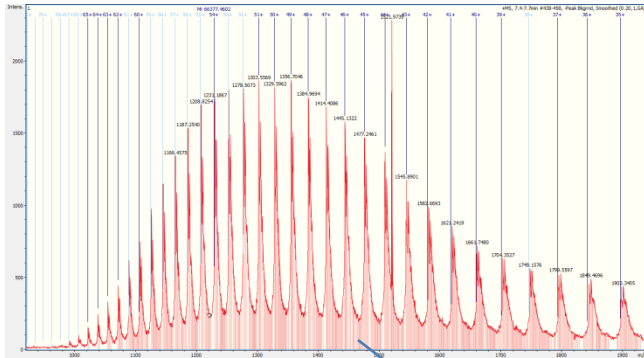
ESI-TOF spectrum of intact BSA



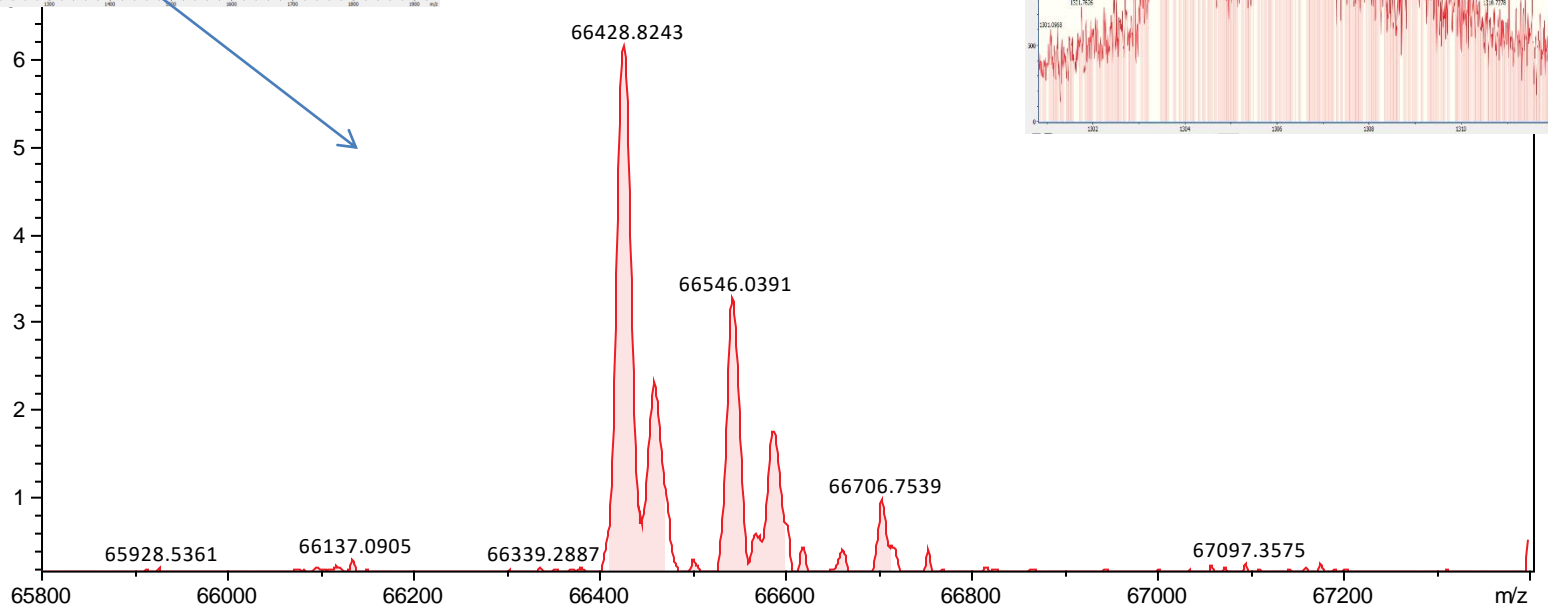
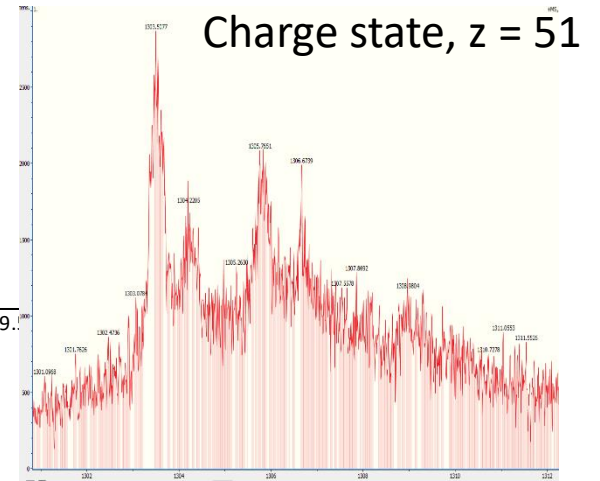
Intact proteins >30 kDa usually determined as average mass



Using deconvolution algorithms



4-7.7min, -Peak Bkgrnd, Deconvoluted (MaxEnt, 1157.73-1499..)

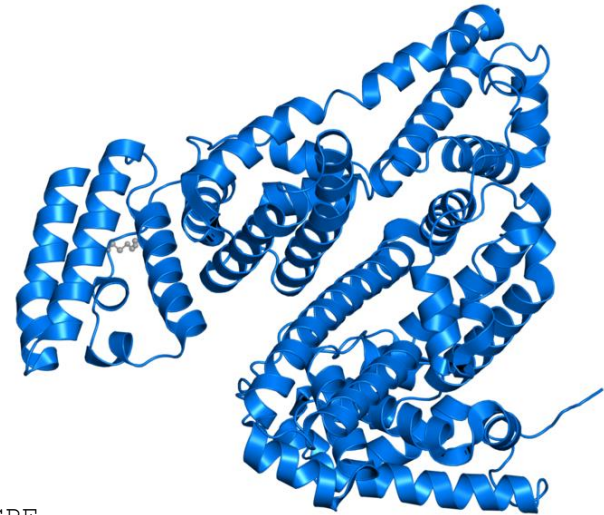


Take-home message

- Mass spectra depict m/z
- Proteins are usually multiply charged and display several charge states in ESI
- Proteins >30 kDa are usually determined as average mass
- Spectrum displaying several charge states needs to be deconvoluted to yield the non-charged average mass of the protein.

When the results do not match...

- ESI-TOF: 66428.8 Da
- Calc.: 69293.4 Da



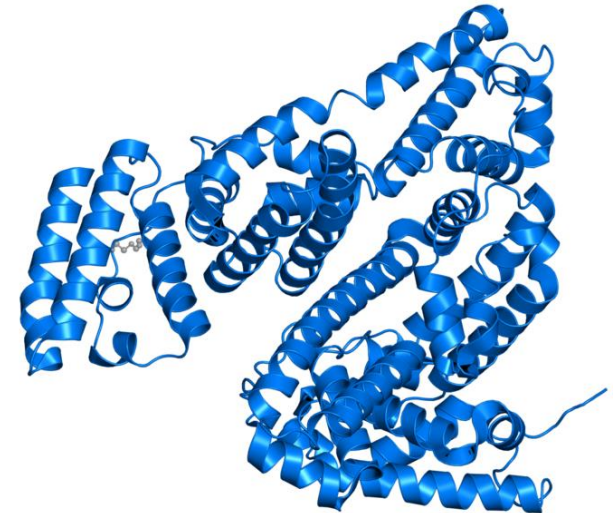
```
>sp|P02769|ALBU_BOVIN Serum albumin
MKWVTFISLLLLFSSAYSRGVFRRDTHKSEIAHRFKDLGEEHFKGLVLIAFSQYLQQCPF
DEHVKLVNELTEFAKTCVADESHAGCEKSLHTLFGDELCKVASLRETYGDMADCCEKQEP
ERNECFLSHKDDSPDLPKLPDPNTLCDEFKADEKKFWGKLYEIAARRHPYFYAPELLYY
ANKYNGVVFQECQAEDKGACLLPKIETMREKVLASSARQRLRCASIQKFGERALKAWSVA
RLSQKFPKAEFVEVTKLVTDLTKVHKECCHGDLLECADDRADLAKYICDNQDTISSKLKE
CCDKPLLEKSHCIAEVEKDAIPENLPPLTADFAEDKDVCKNYQEAKDAFLGSFLYEYSRR
HPEYAVSVLLRLAKEYEATLECCAKDDPHACYSTVFDKCLKHLVDEPQNLIKQNCDFEKL
LGEYGFQNALIVRYTRKVPQVSTPTLVEVSRSLGKVGTRCCTKPESERMPCTEDYLSLIL
NRLCVLHEKTPVSEKVTKCTESLVNRRPCFSALTPDETYVPKAFDEKLFTHADICTLP
DTEKQIKQTALVELLKHKPKATEEQKLTVMENFVAFVDKCCAADDKEACFAVEGPKLVV
STQTALA
```

What is wrong?

Protein processing

>sp|P02769|ALBU_BOVIN Serum albumin

MKWVTFISLLLLFSSAYSRGVFRR DTHKSEIAHRFKDLGEEHFKGLVLIAFSQYLQQCPF
DEHVKLVNELTEFAKTCVADESHAGCEKSLHTLFGDELCKVASLRETYGDMADCCEKQEP
ERNECFLSHKDDSPDLPKLKPDPNTLCDEFKADEKKFWGKYLIEIARRHPYFYAPELLYY
ANKYNGVFQECCQAEDKGACLLPKIETMREKVLASSARQRLRCASIQKFGERALKAWSVA
RLSQKFPKAEFVEVTKLVTDLTKVHKECCHGDLLECADDRADLAKYICDNQDTISSKLKE
CCDKPLLEKSHCIAEVEKDAIPENLPPLTADFAEDKDVCNKYQEAKDAFLGSFLYEYSRR
HPEYAVSVLLRLAKEYEATLEECCAADDPHACYSTVFDKLLHLVDEPQNLIKQNCDOFEK
LGEYGFQNALIVRYTRKVPQVSTPTLVEVSRSLGKVGTRCCTKPESERMPCTEDYLSLIL
NRLCVLHEKTPVSEKVTCCTESLVNRRPCFSALTPDETYVPKAFDEKLFTFHADICTLP
DTEKQIKKQTALVELLKHKPKATEEQKLTVMENFVAFVDKCCAADDKEACFAVEGPKLVV
STQTALA



69293.41 Da full chain

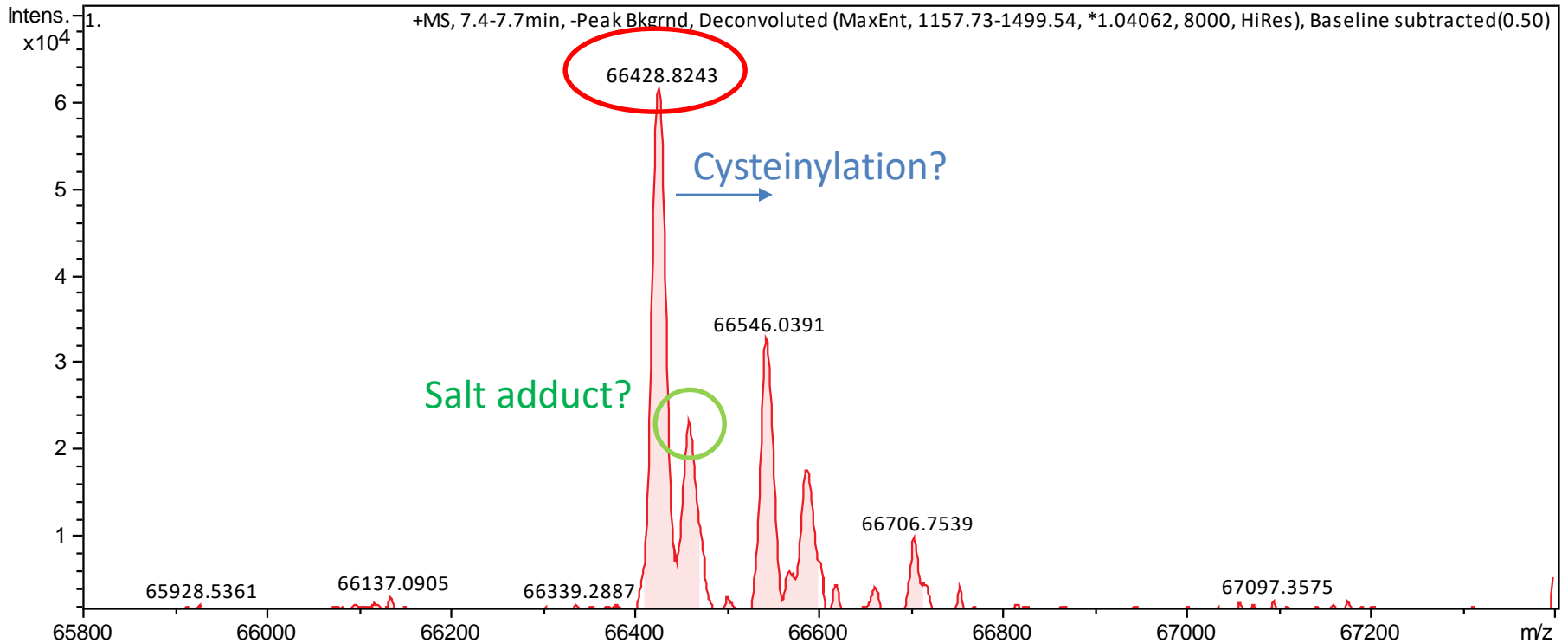
66432.96 Da mature chain (reduced state)

Plus:

- 17 disulfide bonds, 1 free –SH
- 12 other PTMs (Phospho, succinyl)
- Dimerizes
- Binds water, Ca, Na, K, Zn, fatty acids, hormones, drugs, etc.
- Chemical modifications

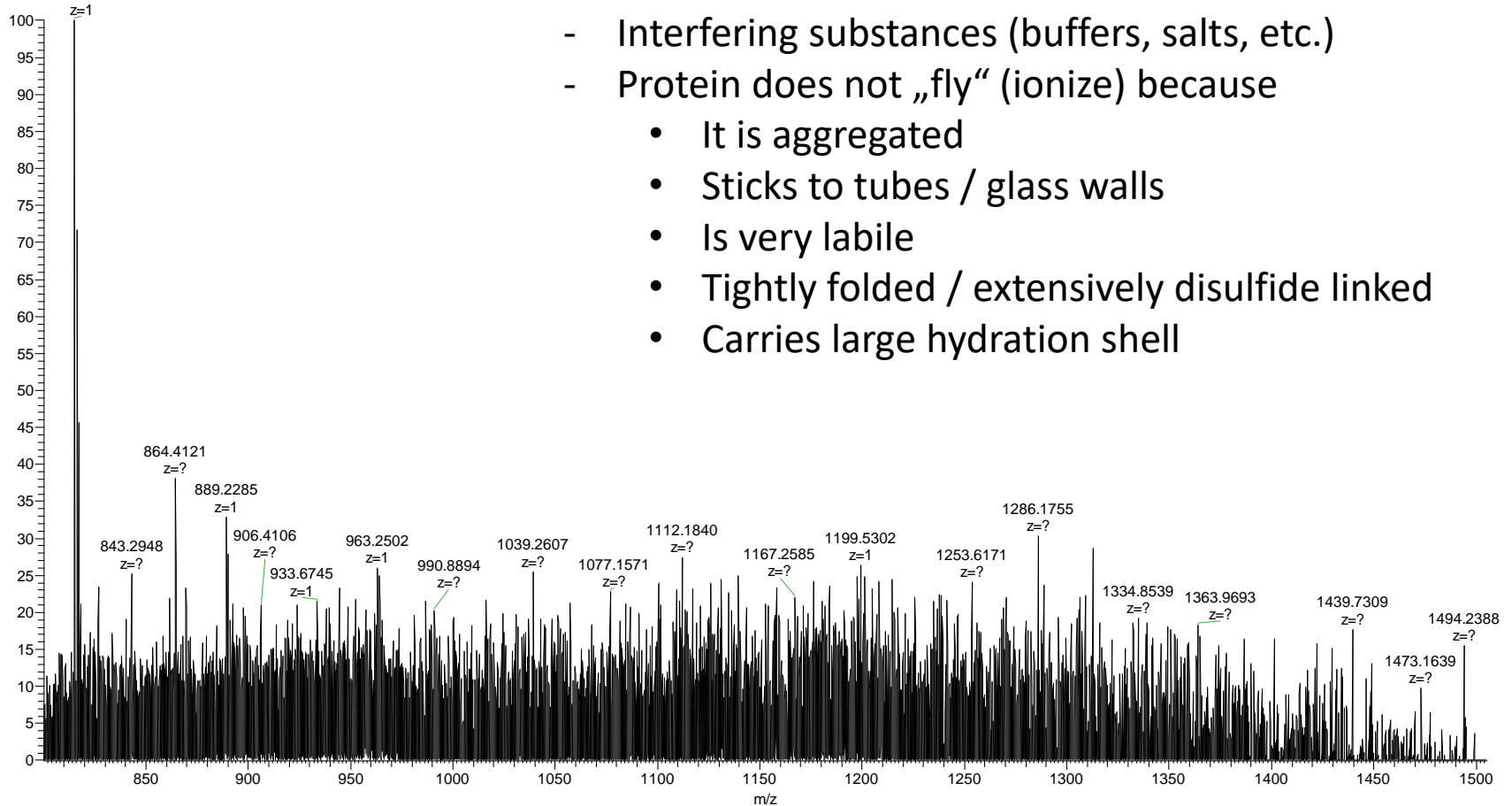
Interpreting the unknowns

-4 Da, not fully reduced (disulfide bonds)?



Sometimes nothing works

20150417_ipaj_ft_60k_01 #1-28 RT: 0.01-1.13 AV: 28 NL: 1.81E4
T: FTMS + p.NSI Full ms [800.00-1500.00]
815.2109



Possible reasons:

- Interfering substances (buffers, salts, etc.)
- Protein does not „fly“ (ionize) because
 - It is aggregated
 - Sticks to tubes / glass walls
 - Is very labile
 - Tightly folded / extensively disulfide linked
 - Carries large hydration shell

The limits of this idea?

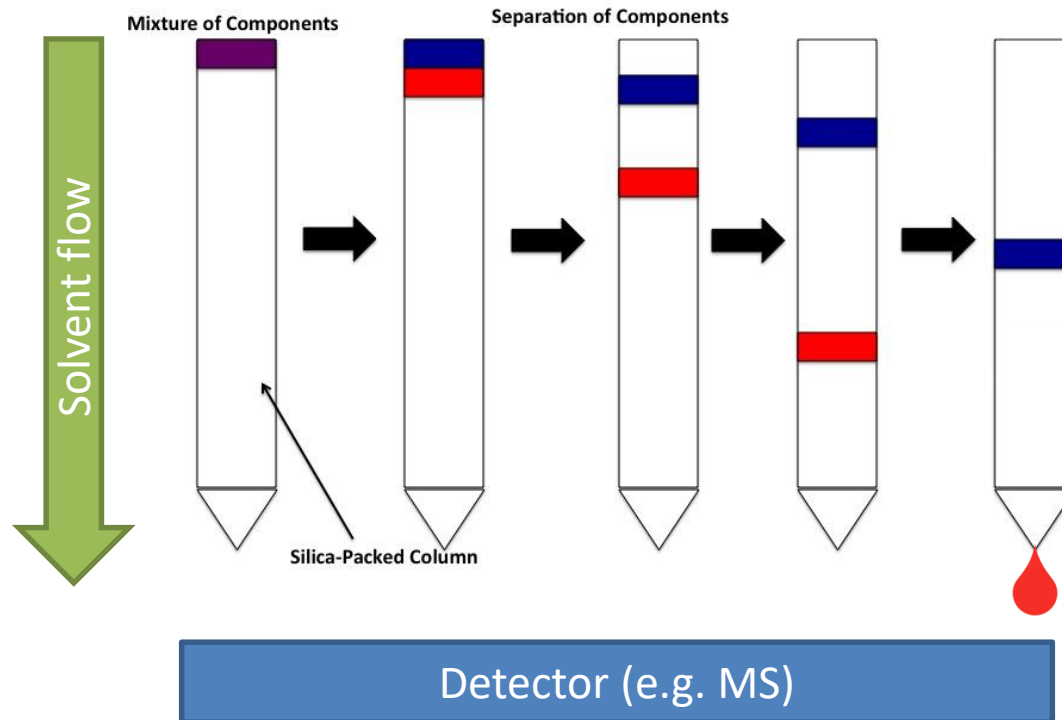


Intact mass analysis limited by:

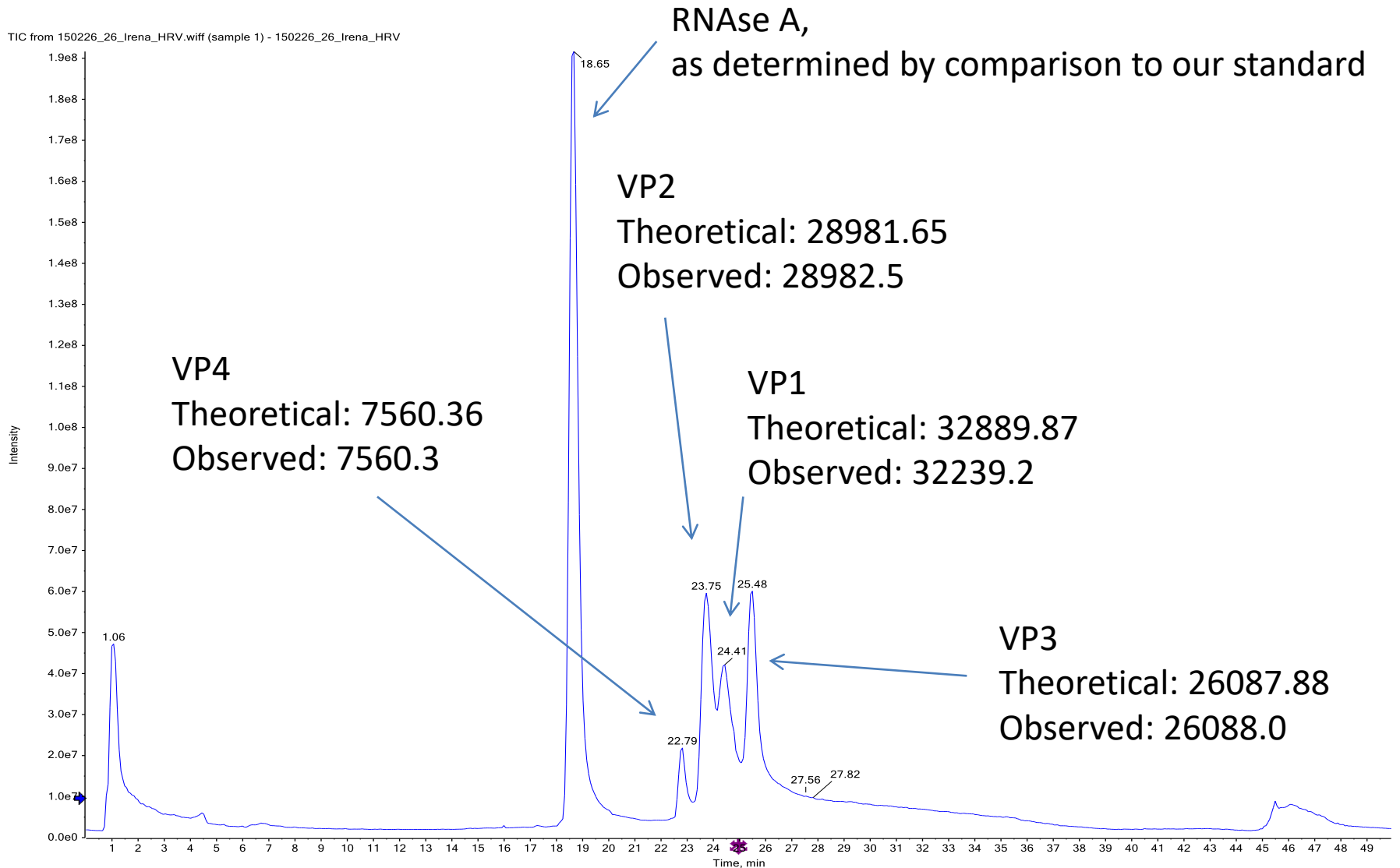
- Complex mixtures of unknown composition
- Analytes with isobaric mass
- Unknown modifications or processing of analytes
- Technical issues with biomolecules

The „LC“ in LC-MS

LC = liquid chromatography



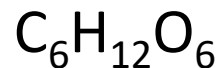
Human rhinovirus



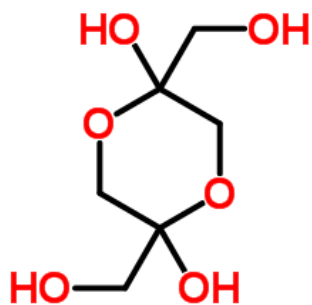
Intact mass analysis limited by:

- Complex mixtures of unknown composition
- Analytes with isobaric mass
- Unknown modifications or processing of analytes
- Technical issues with biomolecules

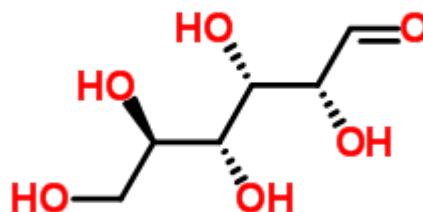
Sometimes one mass is not specific enough



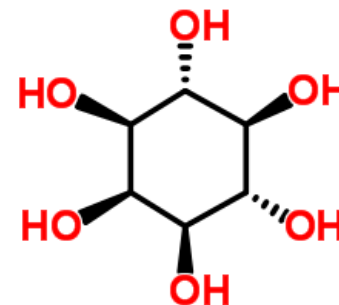
Monoisotopic mass: 180.0634 Da



1,3-Dihydroxyacetone Dimer



D-(+)-Glucose



Inositol

Sometimes one mass is not specific enough

ISGGDALQSCVDR	1320.4 Da
DVCSQLADGGSIR	1320.4 Da
IpSGGDALQSCVDR	1400.4 Da
ISGGDALQpSCVDR	1400.4 Da

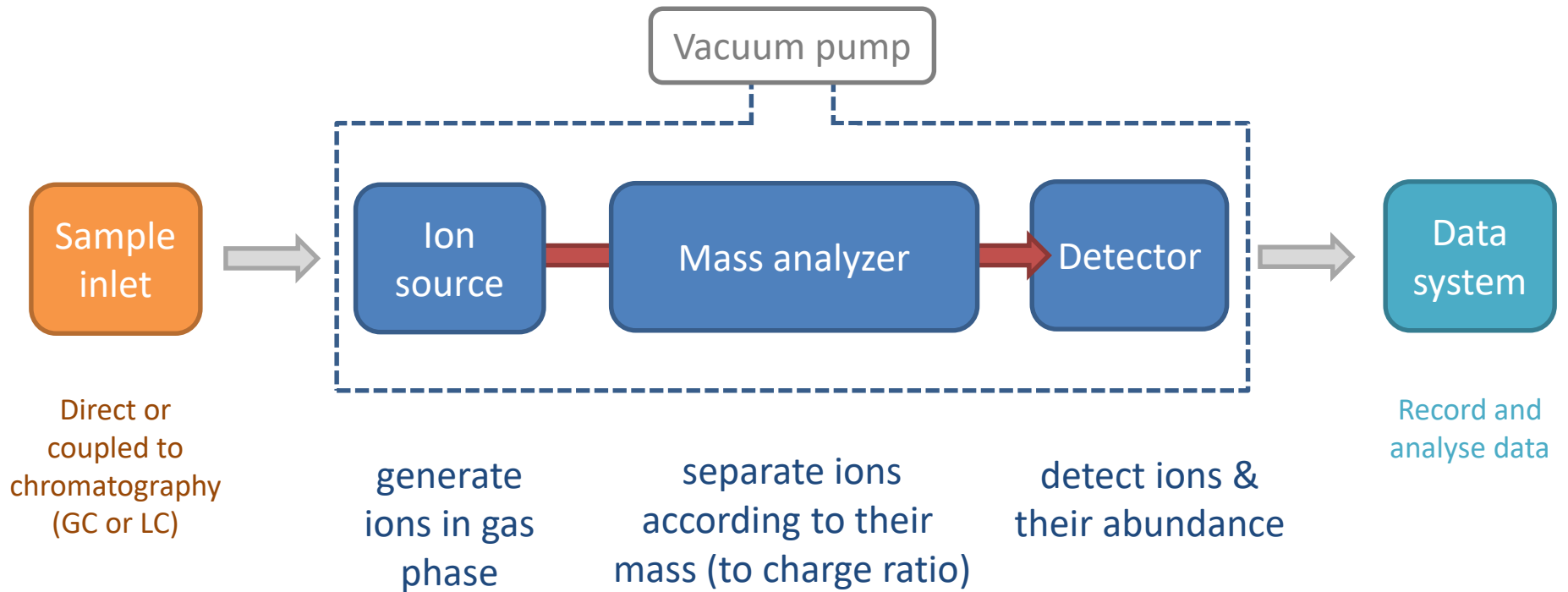
No information about:

- positions of amino acids
- Position of post-translational modifications

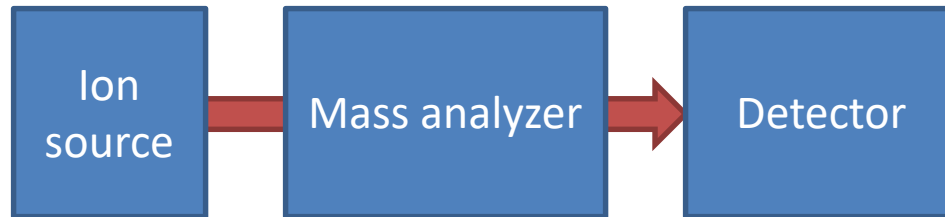
Intact mass analysis limited by:

- Complex mixtures of unknown composition
- Analytes with isobaric mass
- Unknown modifications or processing of analytes
- Technical issues with biomolecules

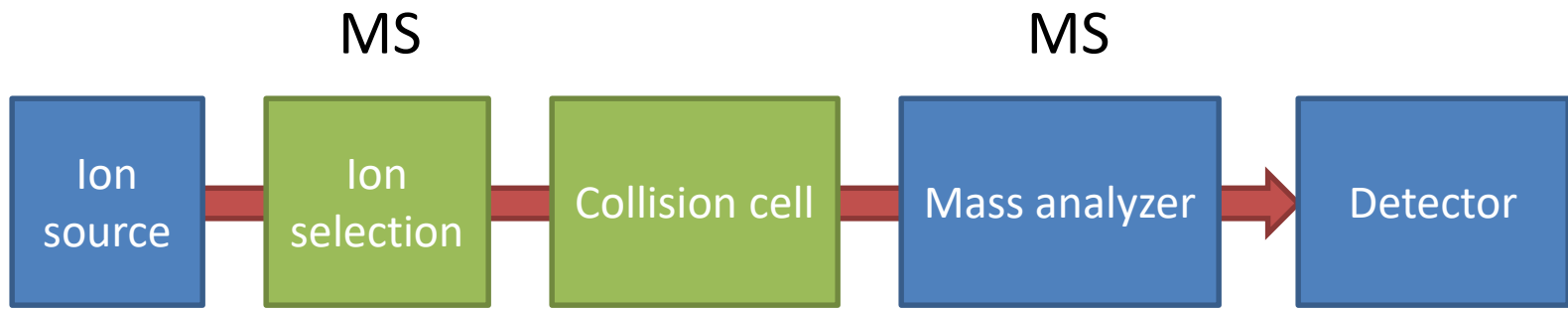
A basic mass spectrometer



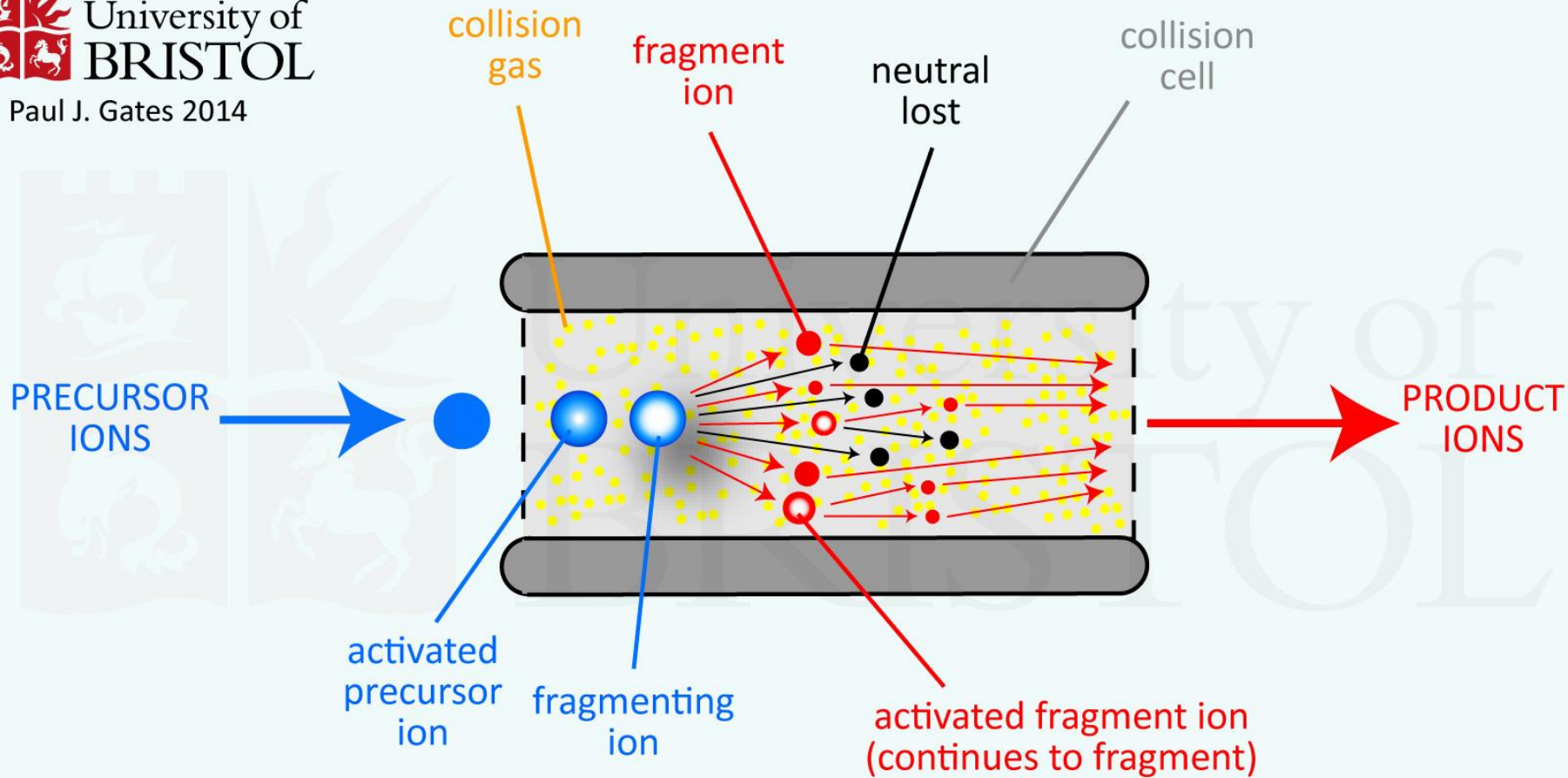
A basic mass spectrometer



Generating more & specific information: The MS/MS approach



A collision cell



Mapping fragments to a sequence

Query	Start - End	Observed	Mr(expt)	Mr(calc)	ppm	M	Score	Expect	Rank	U	Peptide
#2	2 - 41	4631.2100	4630.2027	4630.1915	2.42	0	15	1.2e+002	1	U	M. SHHWGYGKNGPEWHKDFPIANGERQSPVDIDTKAVVQD.P + Acetyl (N-term)
#61	180 - 260	9216.9194	9215.9122	9215.8929	2.09	0	280	4.9e-025	1	U	D. PGSLLPNVLDYWTYPGSLTTPPLLESVTWIVLKEPISVSSQQLKFKRTLNFNAEGEPELLMLANWRPAQPLKNRQVRGFPK.-
#58	185 - 260	8749.6395	8748.6322	8748.6185	1.56	0	268	7.9e-024	1	U	L. PNVLDYWTYPGSLTTPPLLESVTWIVLKEPISVSSQQLKFKRTLNFNAEGEPELLMLANWRPAQPLKNRQVRGFPK.-
#52	189 - 260	8326.3895	8325.3822	8325.3703	1.42	0	240	6.7e-021	1	U	L. DYWTYPGSLTTPPLLESVTWIVLKEPISVSSQQLKFKRTLNFNAEGEPELLMLANWRPAQPLKNRQVRGFPK.-
#49	190 - 260	8211.3695	8210.3622	8210.3434	2.29	0	232	3.5e-020	1	U	D. YWTYPGSLTTPPLLESVTWIVLKEPISVSSQQLKFKRTLNFNAEGEPELLMLANWRPAQPLKNRQVRGFPK.-
#46	191 - 260	8048.2995	8047.2922	8047.2801	1.51	0	228	8.6e-020	1	U	Y. WTYPGSLTTPPLLESVTWIVLKEPISVSSQQLKFKRTLNFNAEGEPELLMLANWRPAQPLKNRQVRGFPK.-
#42	192 - 260	7862.2194	7861.2122	7861.2007	1.45	0	227	7.8e-020	1	U	W. TYPGSLTTPPLLESVTWIVLKEPISVSSQQLKFKRTLNFNAEGEPELLMLANWRPAQPLKNRQVRGFPK.-
#37	193 - 260	7761.1794	7760.1722	7760.1531	2.46	0	222	2.4e-019	1	U	T. YPGSLTTPPLLESVTWIVLKEPISVSSQQLKFKRTLNFNAEGEPELLMLANWRPAQPLKNRQVRGFPK.-
#29	194 - 260	7598.1094	7597.1022	7597.0897	1.64	0	222	2.1e-019	1	U	Y. PGSLLTTPPLLESVTWIVLKEPISVSSQQLKFKRTLNFNAEGEPELLMLANWRPAQPLKNRQVRGFPK.-
#25	195 - 260	7501.0594	7500.0522	7500.0370	2.03	0	204	1.6e-017	1	U	P. GSLTTPPLLESVTWIVLKEPISVSSQQLKFKRTLNFNAEGEPELLMLANWRPAQPLKNRQVRGFPK.-
#22	196 - 260	7444.0295	7443.0222	7443.0155	0.90	0	189	4.3e-016	1	U	G. SLTTPPLLESVTWIVLKEPISVSSQQLKFKRTLNFNAEGEPELLMLANWRPAQPLKNRQVRGFPK.-
#19	197 - 260	7356.9995	7355.9922	7355.9835	1.18	0	183	1.6e-015	1	U	S. LTTTPPLLESVTWIVLKEPISVSSQQLKFKRTLNFNAEGEPELLMLANWRPAQPLKNRQVRGFPK.-
#16	199 - 260	7142.8695	7141.8622	7141.8517	1.46	0	166	8.9e-014	1	U	T. TTPPLLESVTWIVLKEPISVSSQQLKFKRTLNFNAEGEPELLMLANWRPAQPLKNRQVRGFPK.-
#13	200 - 260	7041.8295	7040.8222	7040.8040	2.57	0	160	3.3e-013	1	U	T. PPLLESVTWIVLKEPISVSSQQLKFKRTLNFNAEGEPELLMLANWRPAQPLKNRQVRGFPK.-
#10	210 - 260	5906.1894	5905.1822	5905.1763	0.99	0	87	8.3e-006	1	U	I. VLKEPISVSSQQLKFKRTLNFNAEGEPELLMLANWRPAQPLKNRQVRGFPK.-
#7	213 - 260	5565.9495	5564.9422	5564.9289	2.39	0	70	0.00053	1	U	K. EPISVSSQQLKFKRTLNFNAEGEPELLMLANWRPAQPLKNRQVRGFPK.-
#4	214 - 260	5436.9095	5435.9022	5435.8863	2.92	0	61	0.0037	1	U	E. PISVSSQQLKFKRTLNFNAEGEPELLMLANWRPAQPLKNRQVRGFPK.-

Protein sequence coverage: 46%

Matched peptides shown in **bold red**.

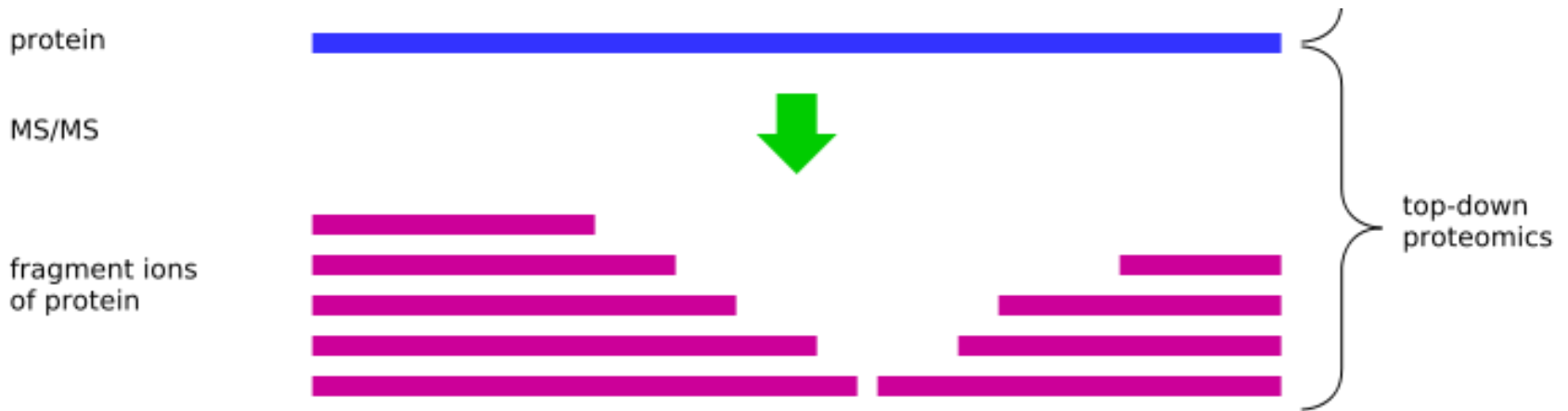
```

1  MSHHWGYGKH NGPEWHKDF PIANGERQSP VDIDTKAVVQ DPALKPLALV
51  YGEATSRRMV NMGHSFNVEY DDSQDKAVLK DGPLTGTYRL VQFHFHWGSS
101 DDQGSSEHTVD RKKYAAELHL VHWNTKYGDF GTAAQQPDGL AVVGVFLKVG
151 DANPALQKVL DALDSIKTKG KSTDFPNFDP GSLLPNVLDY WTYPGSLTTP
201 PLLESVTWIV LKEPISVSSQ QMLKFKRTLNF NAEGEPELLM LANWRPAQPL
251 KNRQVRGFPK

```

Database dependent!

Top-down proteomics



"Bottom-up vs top down" by MagnusPalmlad - Own work. Licensed under CC BY-SA 3.0 via Wikimedia Commons

Drawbacks of top-down proteomics:

- Chromatographic separation of proteins is difficult
- Complex samples are very challenging (charge envelopes, overlapping signals)
- Analysis limited to proteins of 100 kDa max. (for complex mixtures)
- Data analysis algorithms not mature

What's the alternative?

The alternative: break proteins into something more manageable before LC-MS-MS

Digestion of BSA with trypsin: theoretically 75% coverage

```

  10      20      30      40      50      60
dthkseiahr fkDLGEEHFK GLVLIAFSQY LQQCPFDEHV KLVNELTEFA KTCVADESHA

  70      80      90     100     110     120
GCEKSLHTLF GDELCKvasl rETYGMADC CEKqeperNE CFLSHKDDSP DLPKLPDPN

 130     140     150     160     170     180
TLCDEFKade kkfwgkYLYE IARrHPYFYA PELLYYANKY NGVFQECQQA EDKgacllpk

 190     200     210     220     230     240
ietmrekvla ssarqrlrca siqkfgeral kawsvvarlsq kfpkAEFVEV TKLVTDLTKv

 250     260     270     280     290     300
hkECCHGDLLE CADDRadla kYICDNQDTI SSKlkECCDK PLEKSHCIA EVEKDAIPEN

 310     320     330     340     350     360
LPPLTADFAE DKdvckNYQE AKDAFLGSFL YEYSRrHPEY AVSVLLRlak EYEATLEECC

 370     380     390     400     410     420
AKDDPHACYS TVFDKlkHLV DEPQNLIKQN CDQFEKLGEY GFQNALIVRy trkVPQVSTP

 430     440     450     460     470     480
TLVEVSRslg kvgtrCCTKP ESERMPCTED YLSLILNRLC VLHEKtpvse kvtkCCTESL

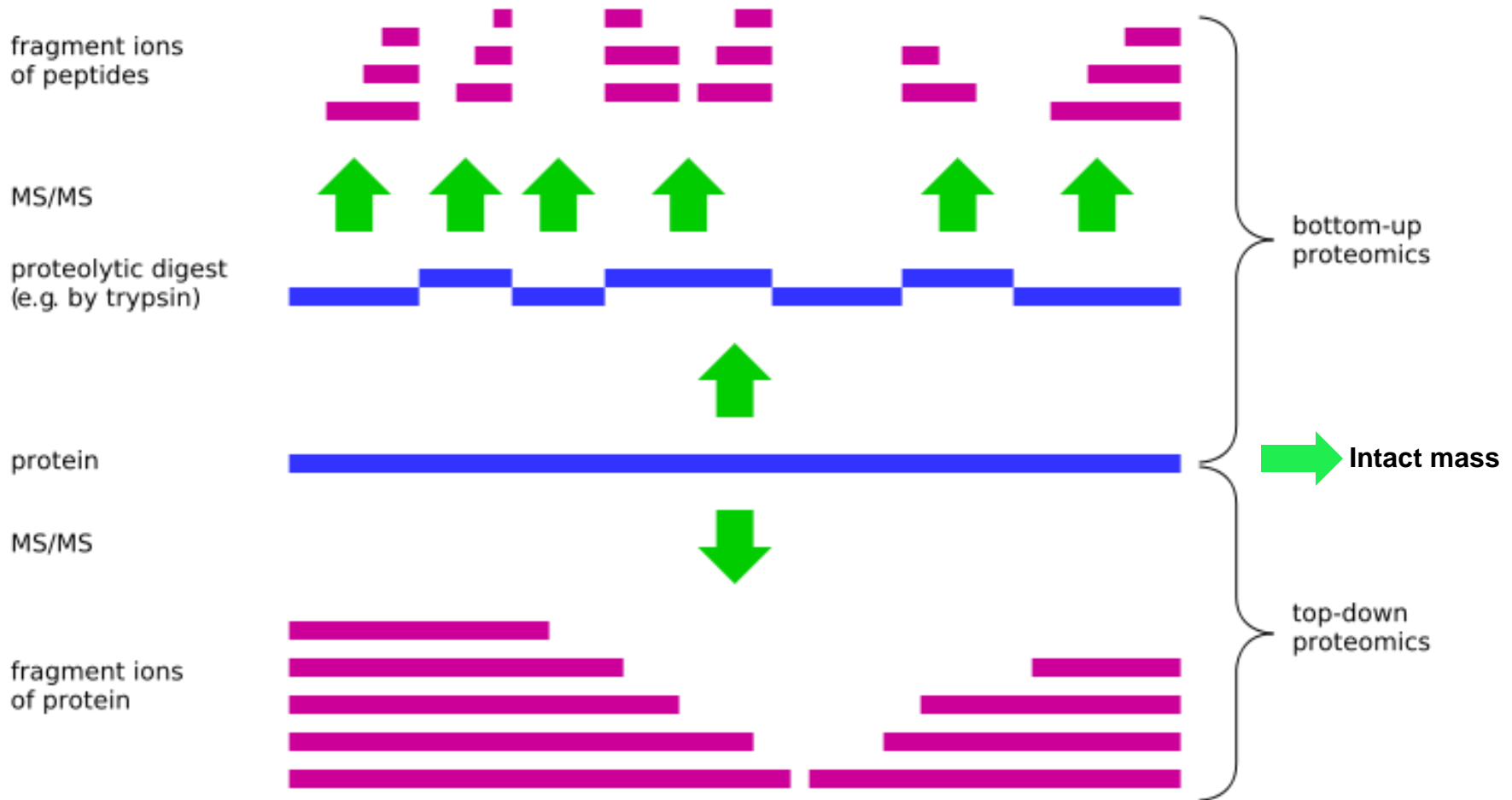
 490     500     510     520     530     540
VNRRPCFSAL TPDETYVPKa fdekLFTFHA DICTLPDTEK qikkQTALVE LLKhkpkATE

 550     560     570     580
EQLKTMENF VAFVDKccaa ddkEACFAVE GPKLVVSTQT ALA
```

Where does trypsin cut?

Why not 100% coverage?

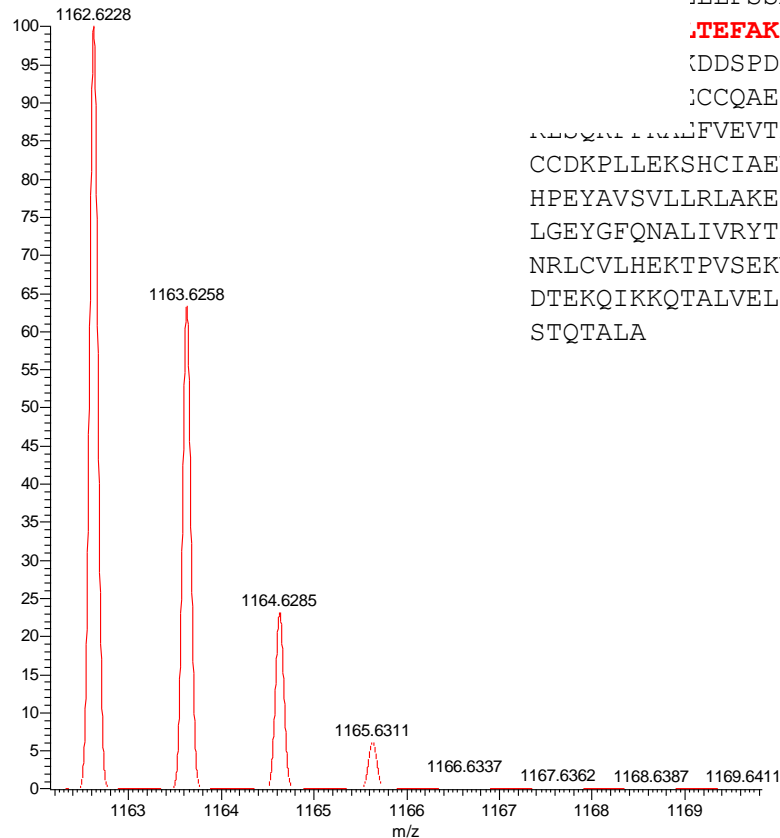
Bottom-up vs. top-down proteomics



The advantage of bottom-up

- Easier chromatography and handling of peptides
- Simpler signals, with accurate monoisotopic masses:

LVNELTEFAK +H2O: C53 H86 N12 O17 p(gss, s/p:40) Chrg...



>sp|P02769|ALBU_BOVIN Serum albumin

```
..LLFSSAYSRGVFRRDTHKSEIAHRFKDLGEEHFKGLVLIAFSQYLQQCPF  
..LVNELTEFAKTCVADESHAGCEKSLHTLFGDELCKVASLRETYGDMADCCEKQEP  
..KDDSPDLPKLKPDPNTLCDEFKADKKFWGKYLEIARRHPYFYAPELLYY  
..KCCQAEDKGACLLPKIETMREKVLASSARQRLRCASIQKFGERALKAWSVA  
..KDDQKPLLEKSHCIAEVEKDAIPENLPPLTADFAEDKDVCKNYQEAKDAFLGSFLYEYSRR  
..HPEYAVSVLLRLAKEYEATLEECCAADDPHACYSTVFDKHLVDEPQNLKQNCDFEK  
..LGEYGFQNALIVRYTRKVPQVSTPTLVEVSRSLGKVGTRCCTKPESERMPCTEDYLSLIL  
..NRLCVLHEKTPVSEKVTKCCTESLVNRRPCFSALTPDETYVPKAFDEKLFTHADICTLP  
..DTEKQIKKQTALVELLKHKPKATEEQKTKVMENFVAFVDKCAADDKEACFAVEGPKLVV  
..STQ TALA
```

Monoisotopic mass: 1162.6228 u

Average mass: 1163.34 u

Nominal mass: 1163 u

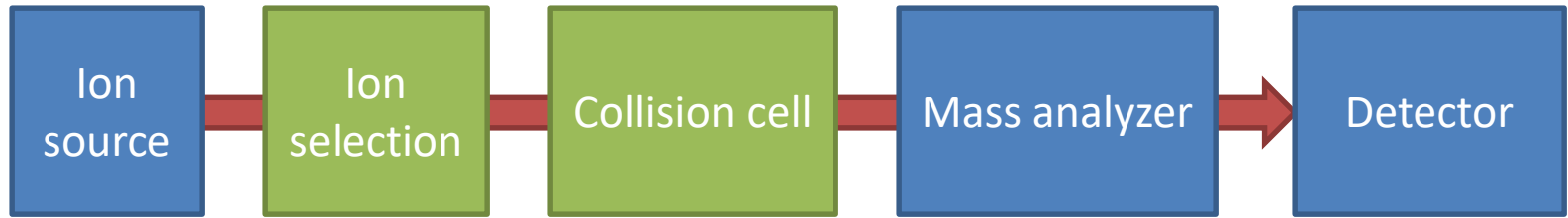
Peptide mass is not enough

ISGGDALQSCVDR	1320.4 Da
DVCSQLADGGSIR	1320.4 Da
IpSGGDALQSCVDR	1400.4 Da
ISGGDALQpSCVDR	1400.4 Da

No information about:

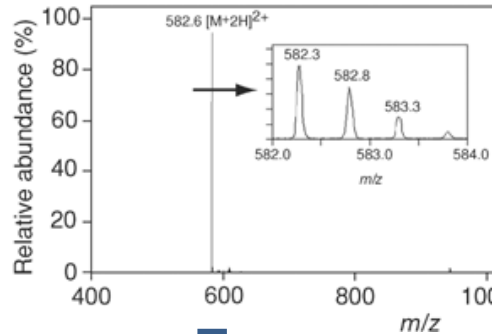
- positions of amino acids
- Position of post-translational modifications

More & specific info: the MS/MS approach

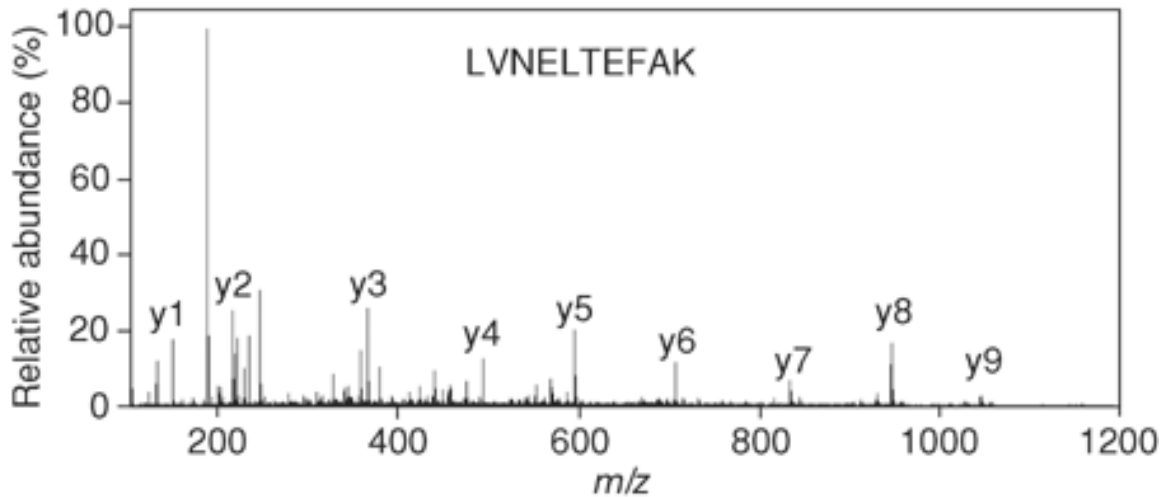


How to identify a peptide from spectrum

MS1 precursor mass

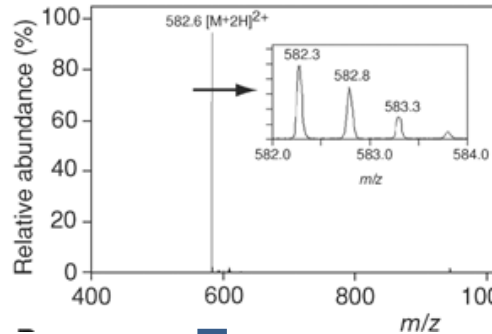


MS/MS fragment masses

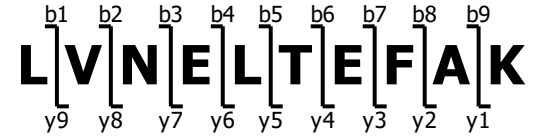
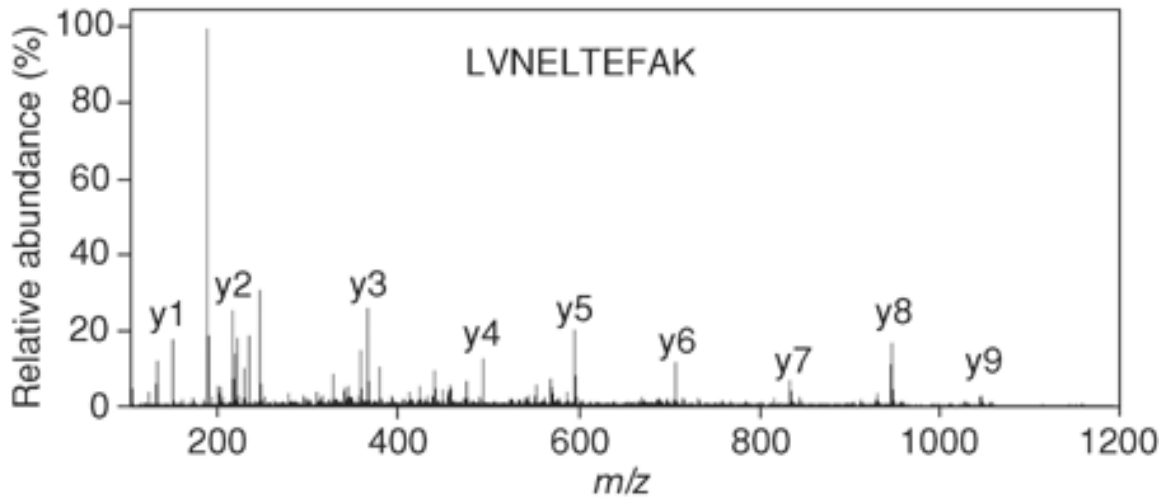


How to identify a peptide from MS/MS spectrum

MS1 precursor mass

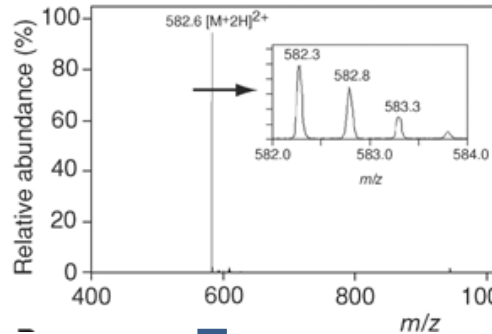


MS/MS fragment masses

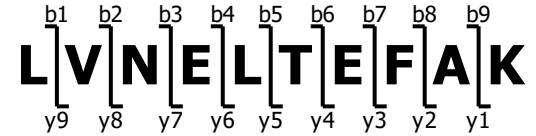
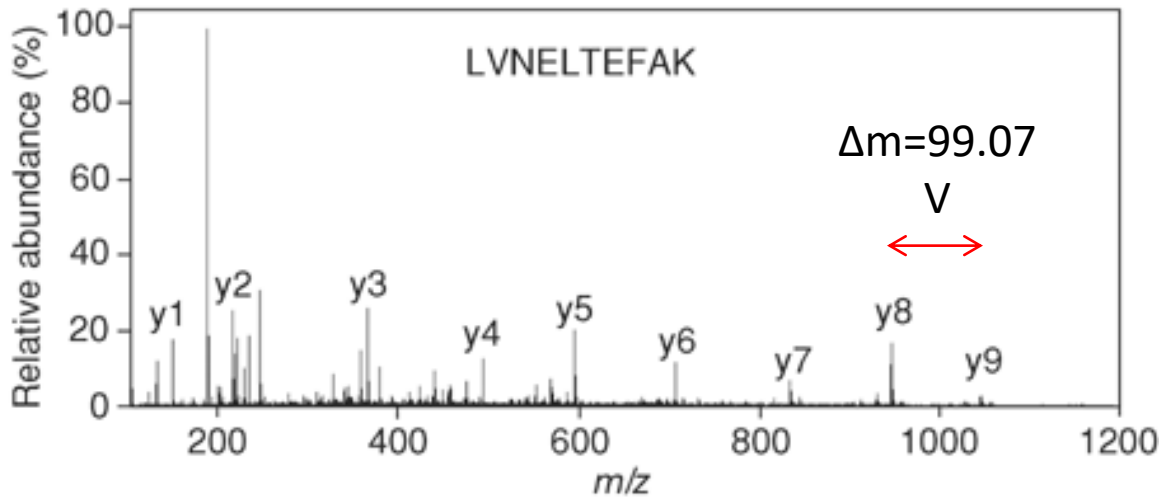


How to identify a peptide from MS/MS spectrum

MS1 precursor mass

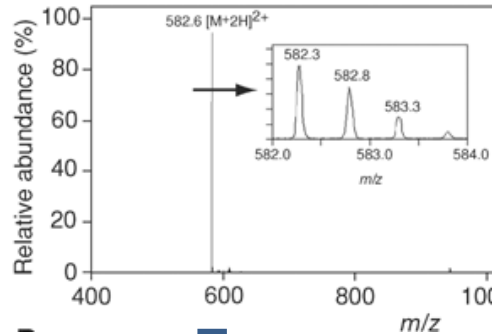


MS/MS fragment masses

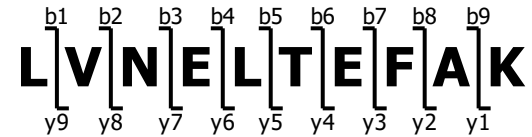
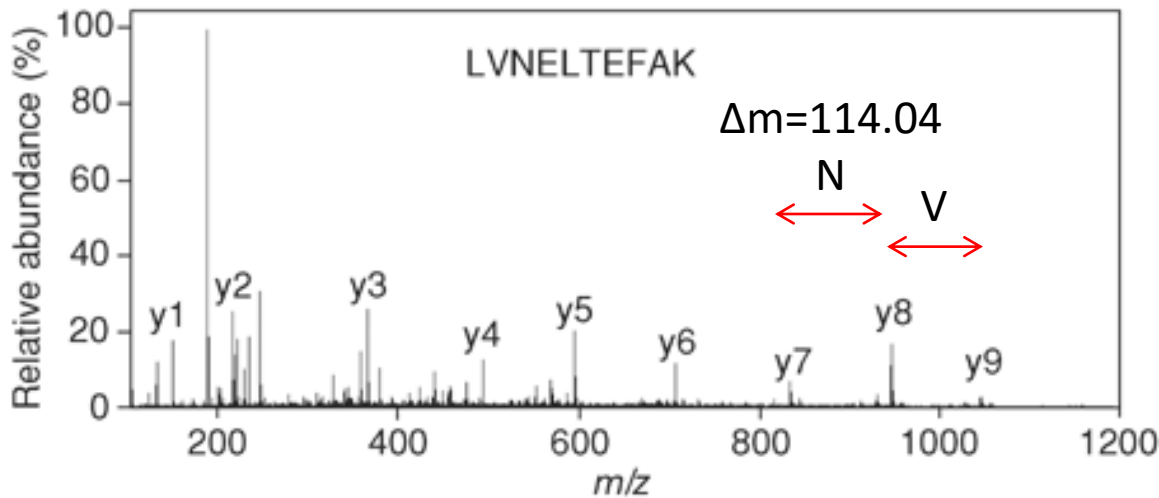


How to identify a peptide from MS/MS spectrum

MS1 precursor mass

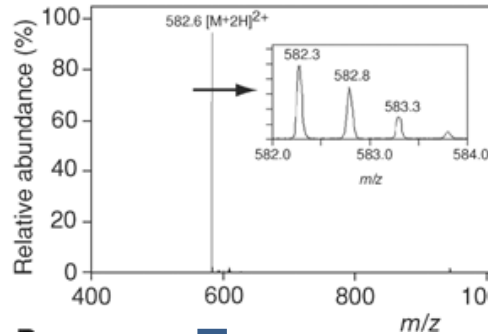


MS/MS fragment masses

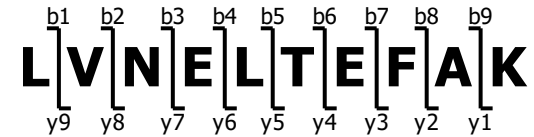
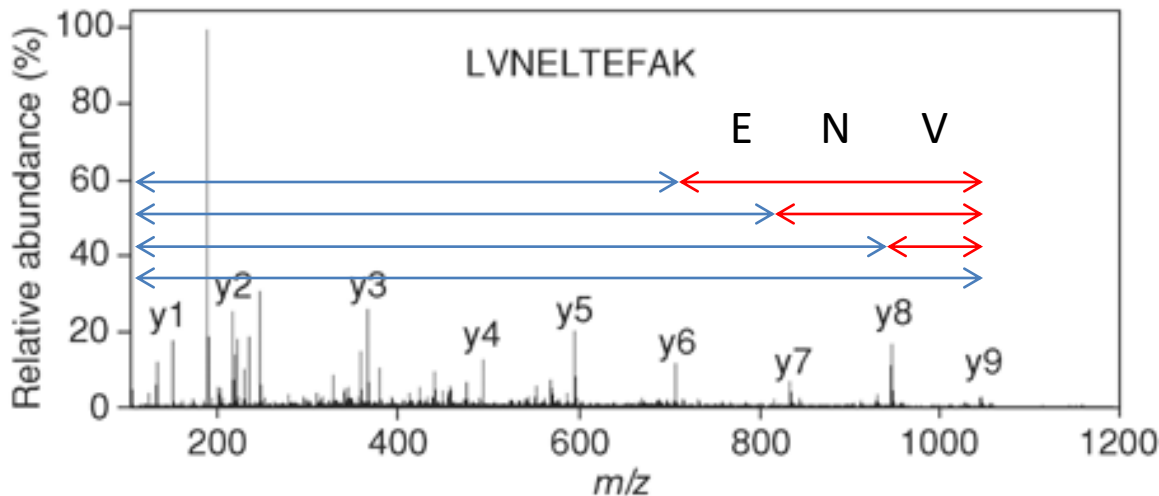


How to identify a peptide from MS/MS spectrum

MS1 precursor mass



MS/MS fragment masses



b	Res.	y
114.092	1 L 10	
213.160	2 V 9	1050.547
327.203	3 N 8	951.478
456.246	4 E 7	837.435
569.330	5 L 6	708.393
670.378	6 T 5	595.309
799.420	7 E 4	494.261
946.489	8 F 3	365.218
1017.526	9 A 2	218.150
	10 K 1	147.113

Information:
Precursor mass
Ion-series



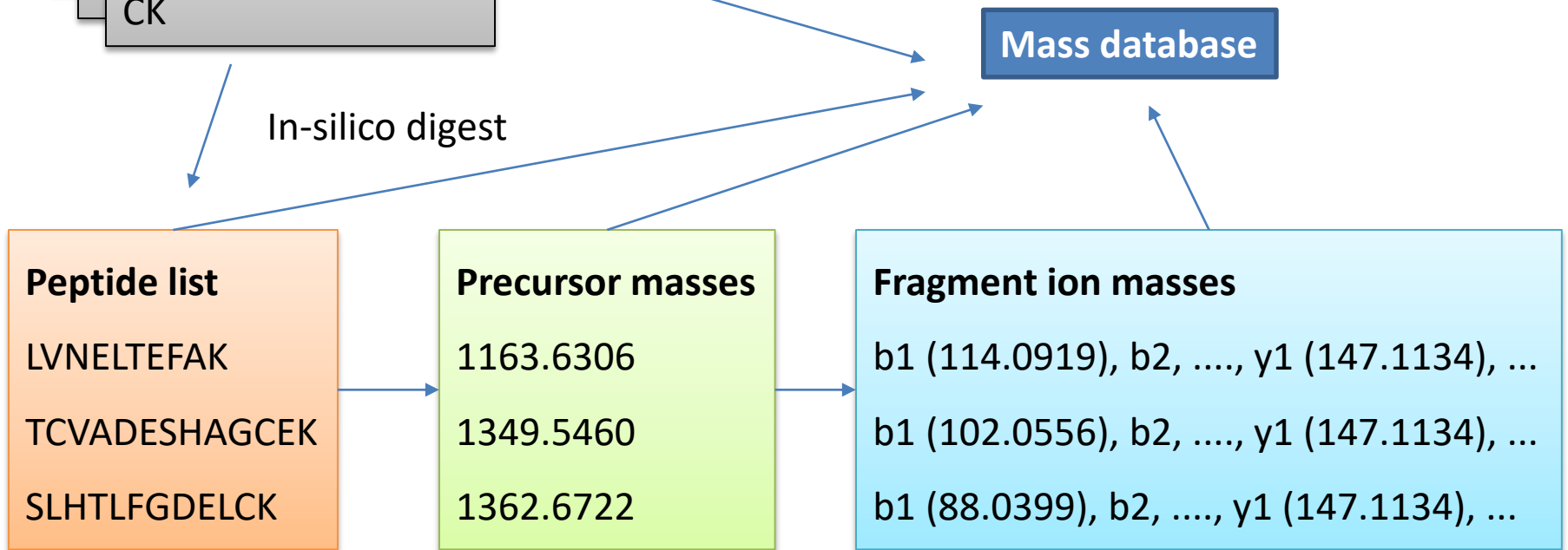
De-novo sequencing or
comparison with database

The basic idea of a DB searches

Database of protein sequences:



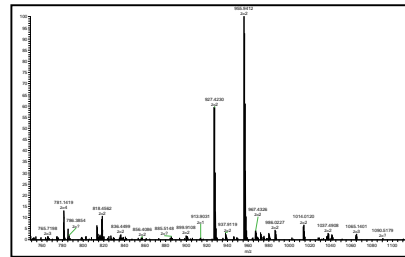
Take a sequence database and calculate masses of all tryptic peptides and their fragment ions to generate a mass database.



The basic idea of a DB searches

Acquired raw data

MS 1: precursor masses



Coverage?

How many proteins can you detect?

In a 2 h run on the newest generation instrument:

- 25.000 peptides (>50.000 spectra)
- >4.000 proteins (depending on organism)
- at 1% FDR!

Sample requirement: 2 μ g peptide sample

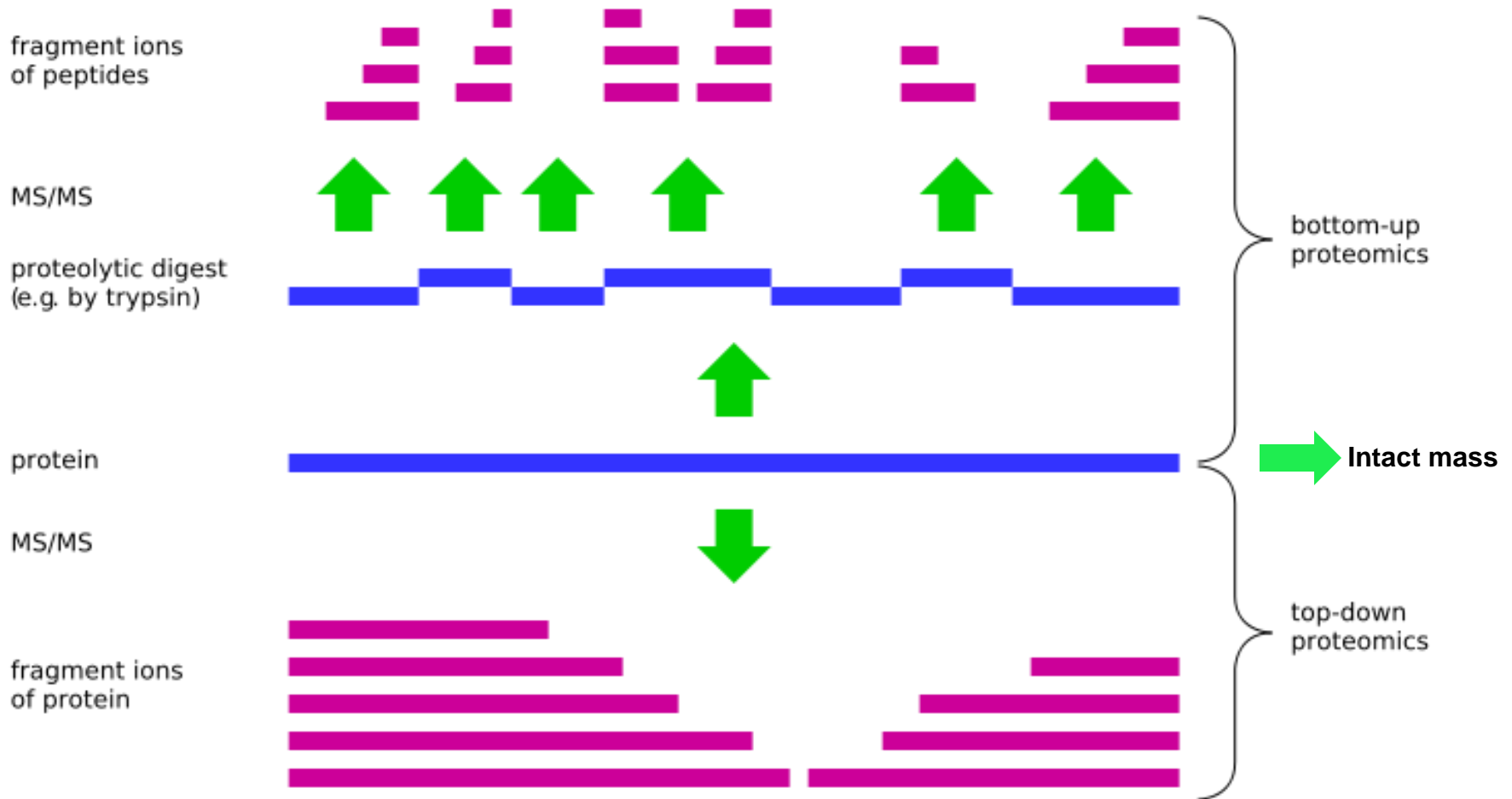
Sample preparation time: 6h plus digest time

Coverage at protein level?

- Almost never 100%, typically 1-80%
- Modifications with low stoichiometry lost

```
      10      20      30      40      50      60
dthkseiahr fkDLGEEHFK GLVLIAFSQY LQQCPFDEHV KLVNELTEFA KTCVADESHA
      70      80      90     100     110     120
GCEKSLHTLF GDELCKvasl rETYGMADC CEKqeperNE CFLSHKDDSP DLPKLPDPN
      130     140     150     160     170     180
TLCDEFKade kkfwgkYLYE IARrHPYFYA PELLYYANKY NGVFQECQA EDKgacllpk
      190     200     210     220     230     240
ietmrekvla ssarqrlrca siqkfgeral kawsvlarlsq kfpkAEFVEV TKLVTDLTKv
      250     260     270     280     290     300
hkECCHGDLL ECADDRadla KYICDNQDTI SSKlkECCDK PLEKSHCIA EVEKDAIPEN
      310     320     330     340     350     360
LPPLTADFAE DKdvckNYQE AKDAFLGSFL YEYSRrHPEY AVSVLLRlak EYEATLEECC
      370     380     390     400     410     420
AKDDPHACYS TVFDKlkHLV DEPQNLIKQN CDQFEKLGEY GFQNALIVRY trkVPQVSTP
      430     440     450     460     470     480
TLVEVSRslg kvgttrCCTKP ESERMPCTED YLSLILNRLC VLHEKtpvse kvtkCCTESL
      490     500     510     520     530     540
VNRRCFSAL TPDETYVPKa fdekLFTFHA DICTLPDTEK qikkQTALVE LLKhkpkATE
      550     560     570     580
EQLKTMENF VAFVDKccaa ddkEACFAVE GPKLVVSTQT ALA
```

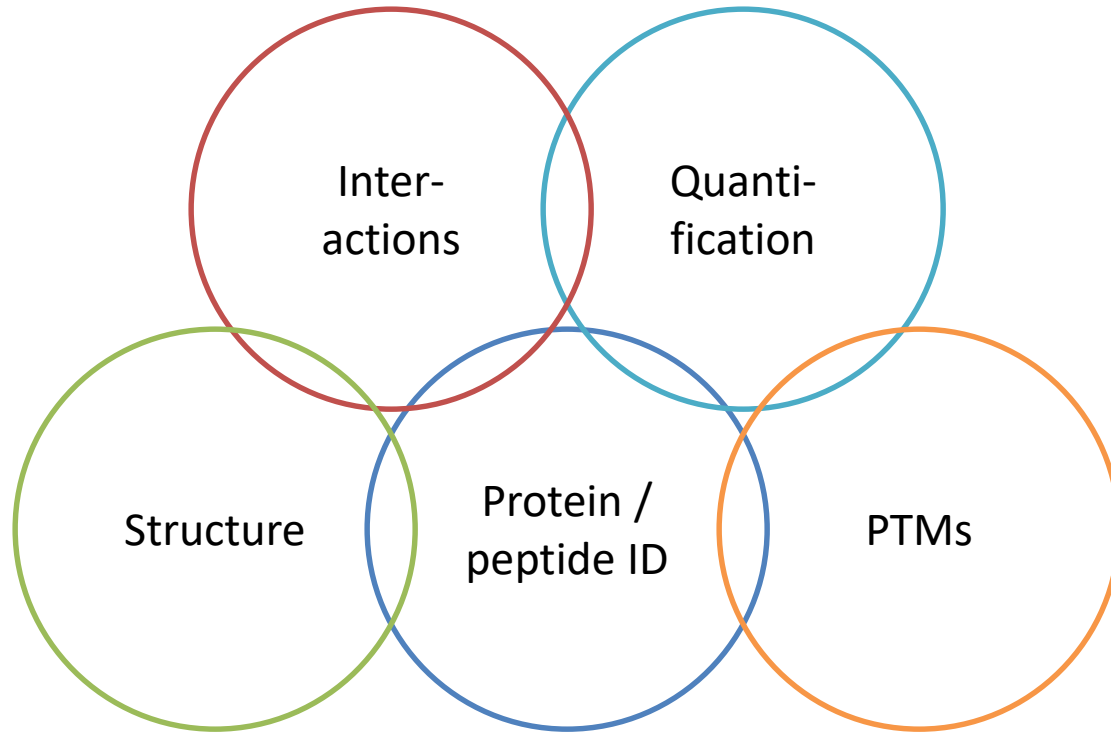
Bottom-up vs. top-down proteomics vs. intact mass analysis



Part II

What is this all useful for?

Applications in proteomics



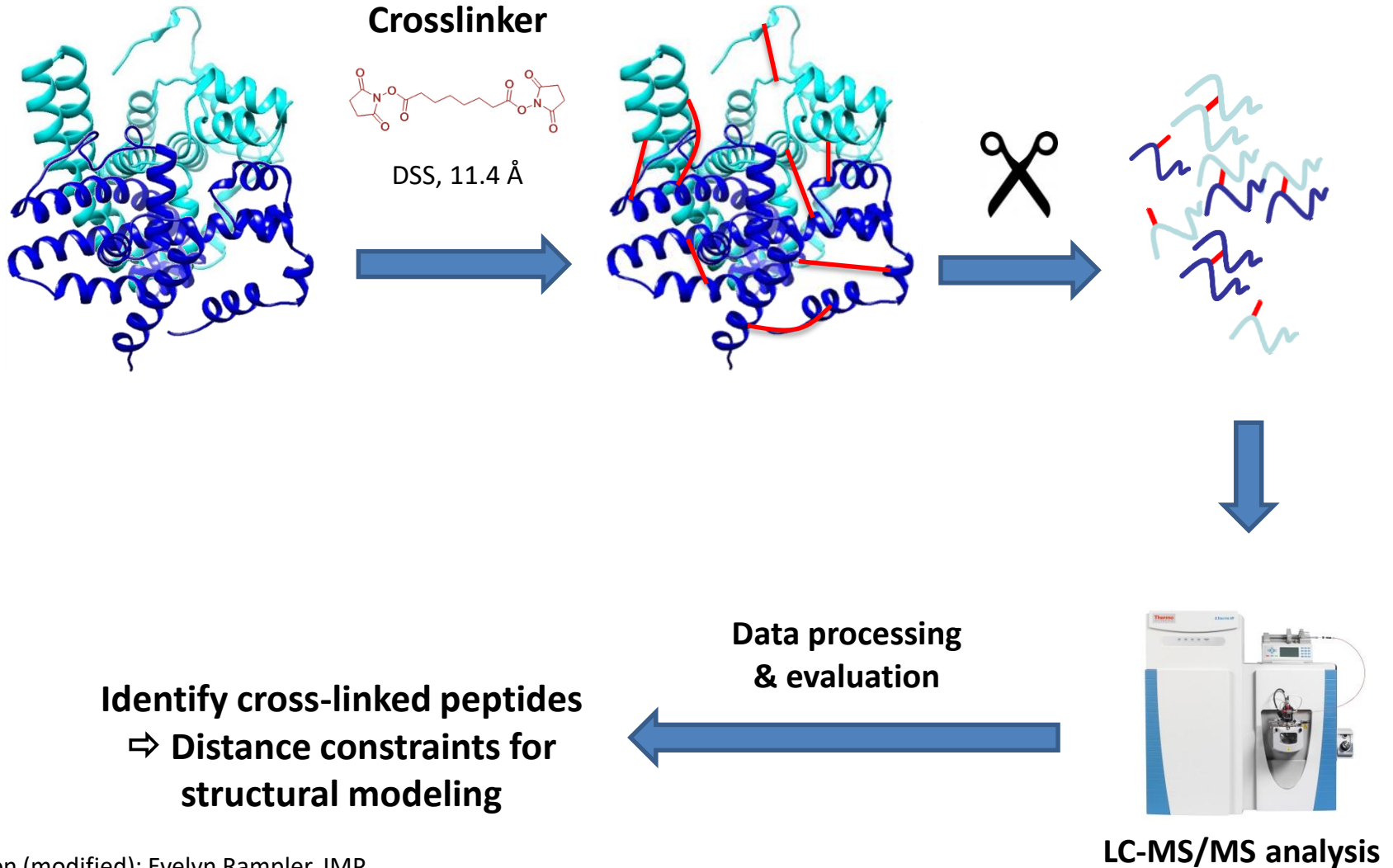
Characterizing structure & interactions with MS

- Identify interaction partners (e.g. Affinity purification MS, BioID, etc.)
- Crosslinking MS
- Hydrogen-deuterium exchange (HDX)
- Native mass spectrometry

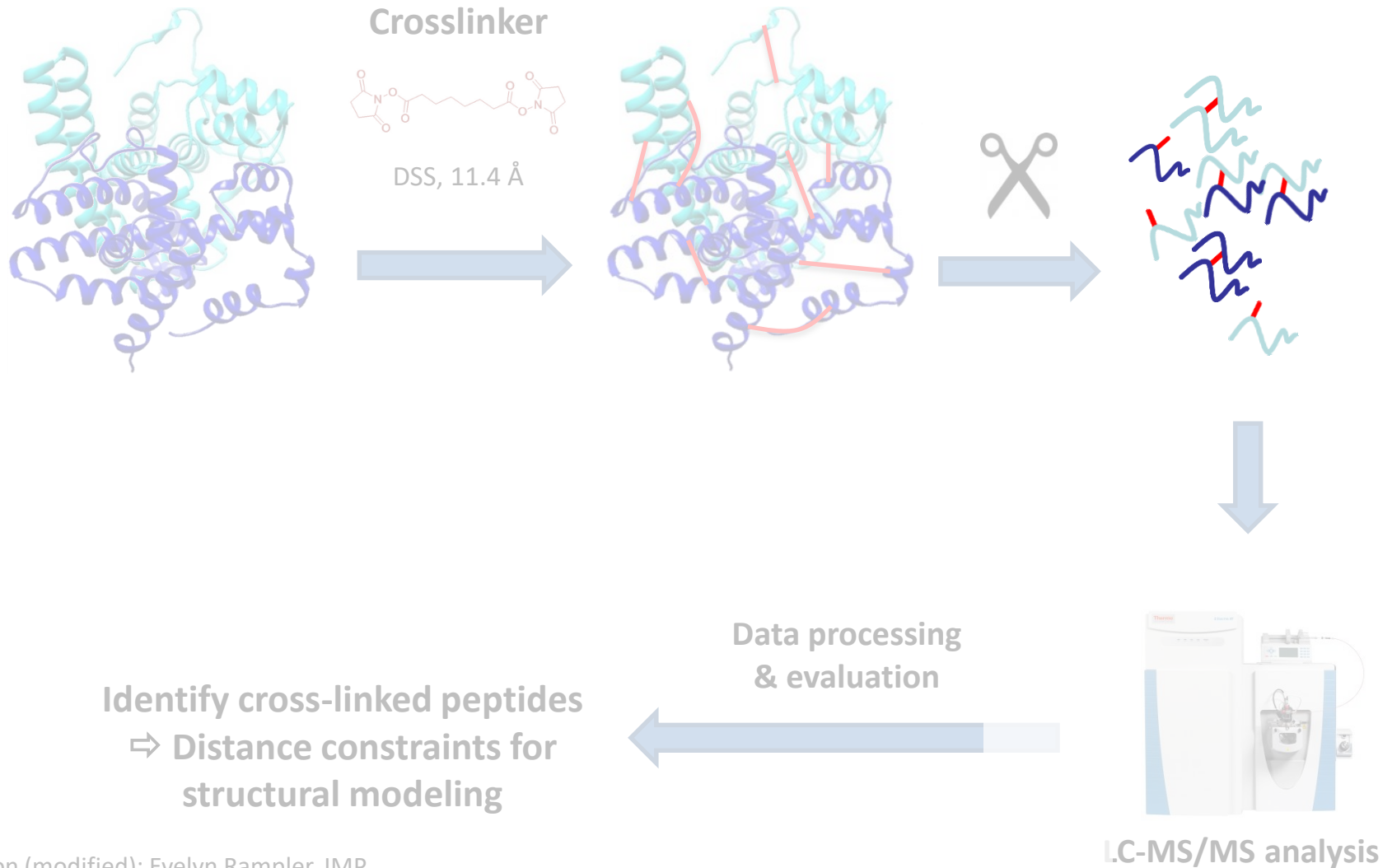
Why crosslinking?

- High-resolution structural tools not always applicable, especially for higher order protein complexes or flexible regions (e.g. IDPs)
- XL-MS is a complementary low-resolution tool
- XL-MS monitors proteins in solution
- What you can gain:
 - Distance constraints
 - Identification of interaction partners
 - Protein complex and network analysis

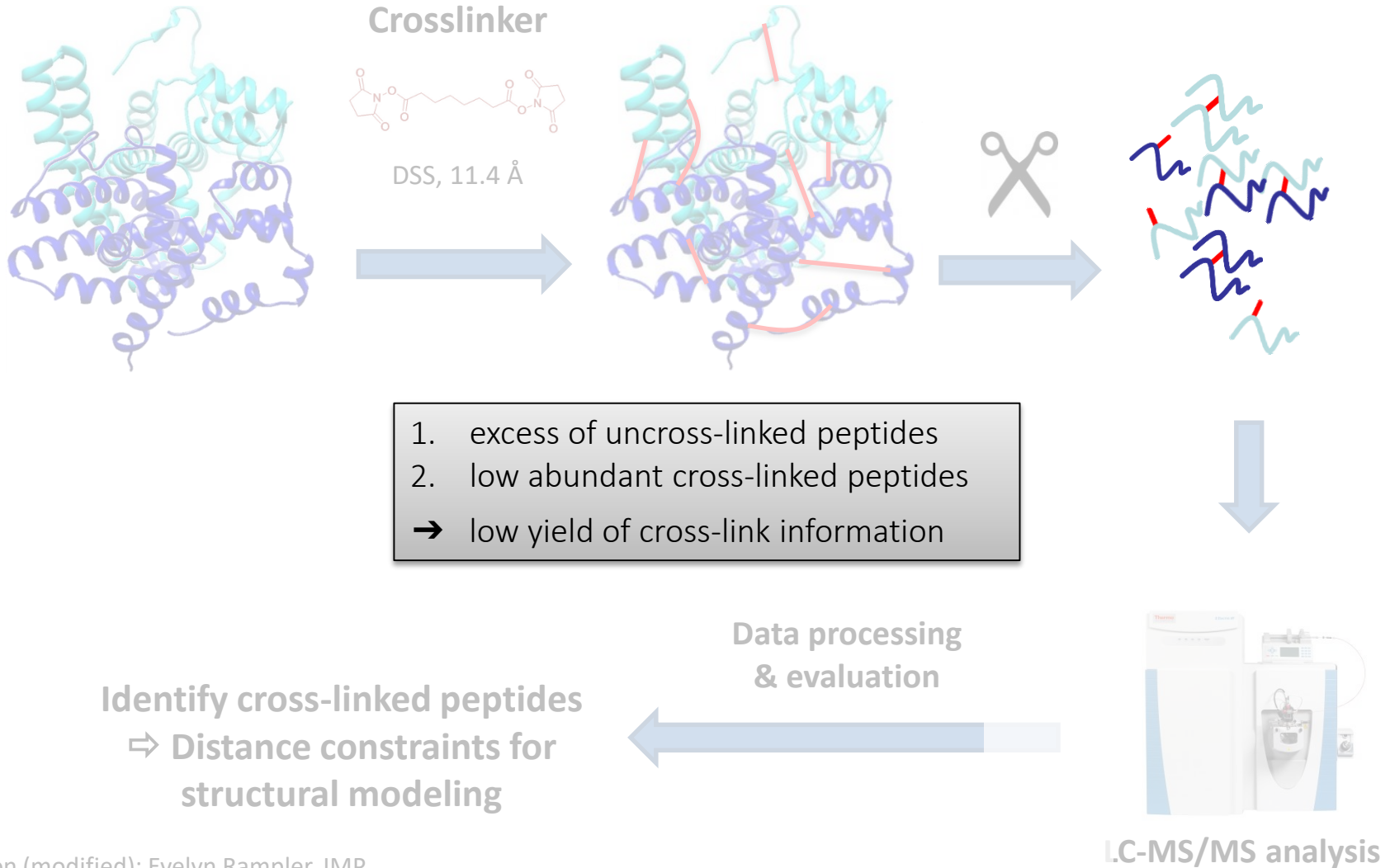
The principle of XL-MS



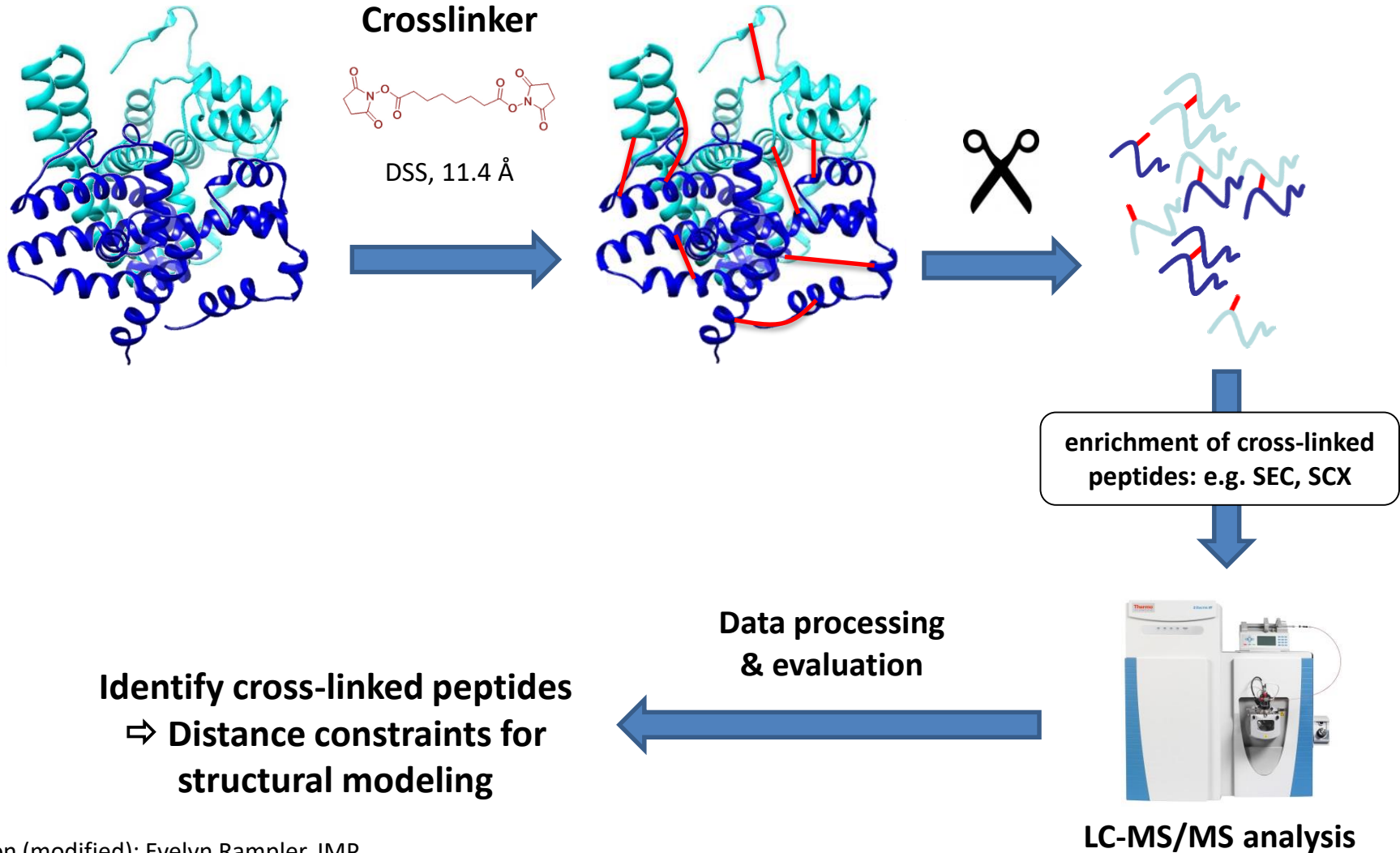
The principle of XL-MS



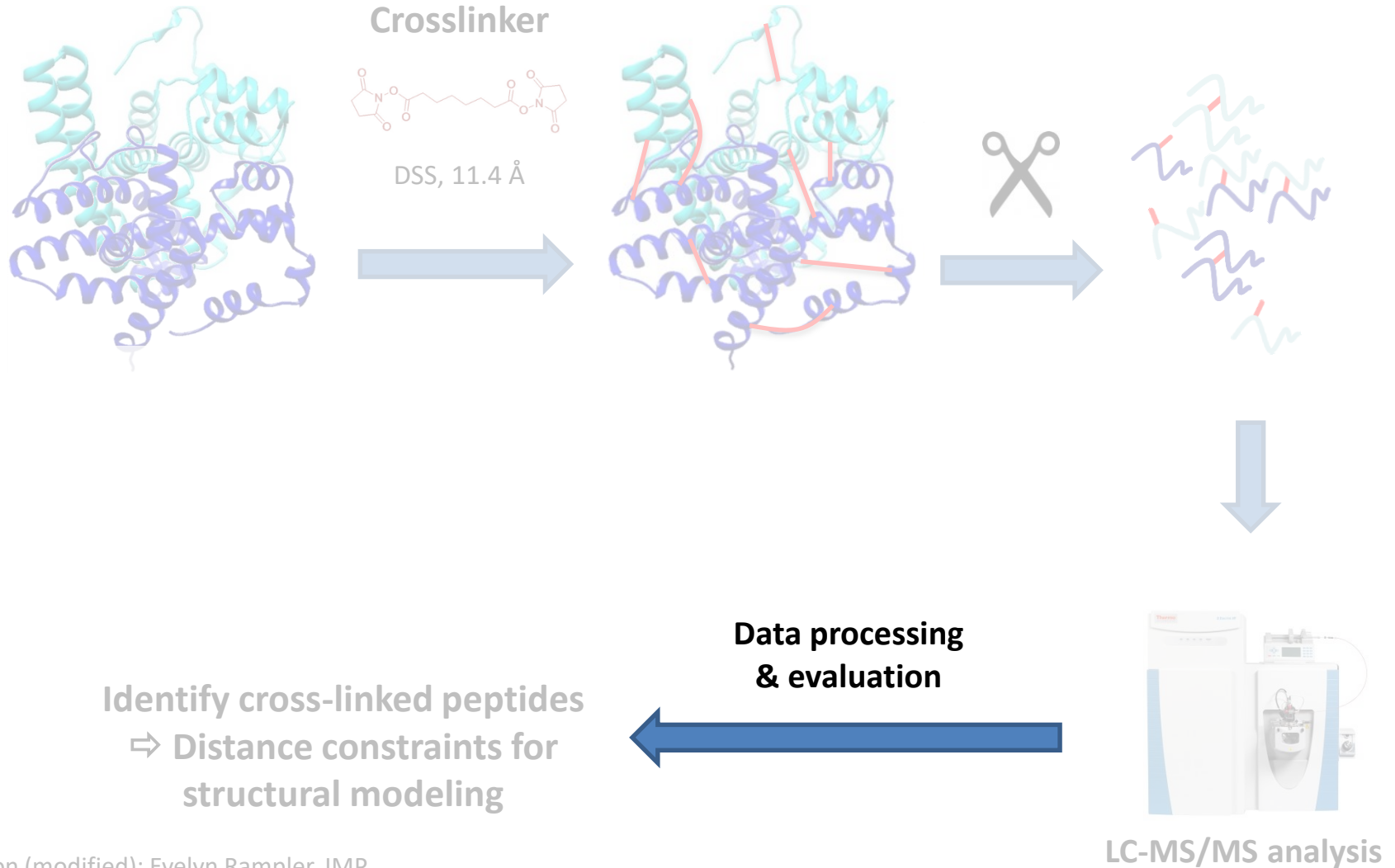
The principle of XL-MS



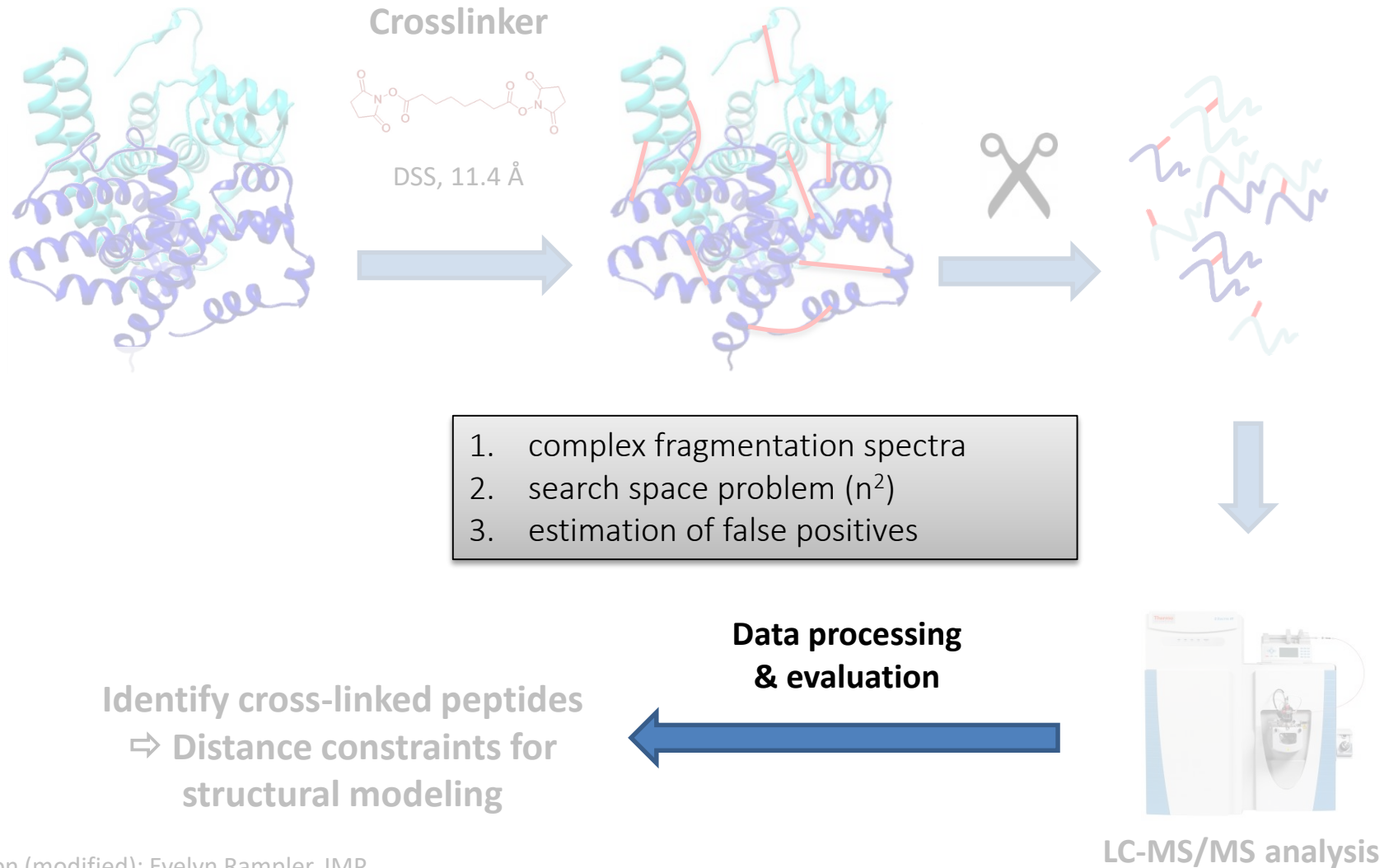
The principle of XL-MS



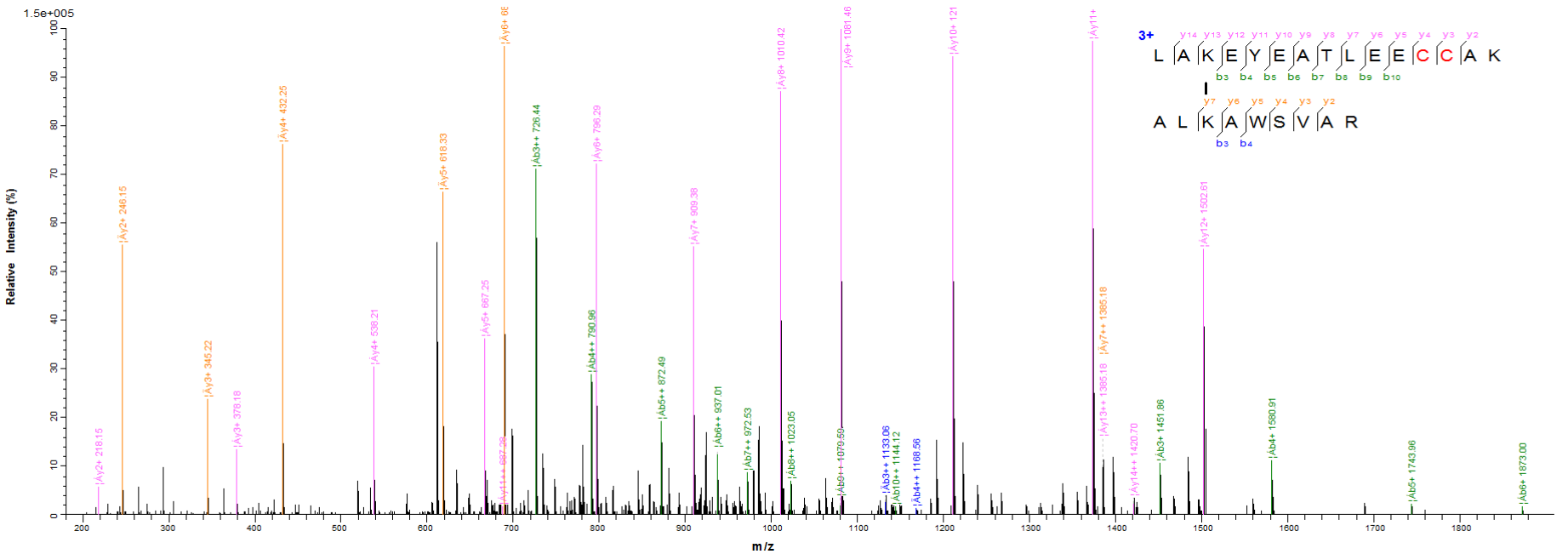
The principle of XL-MS



The principle of XL-MS



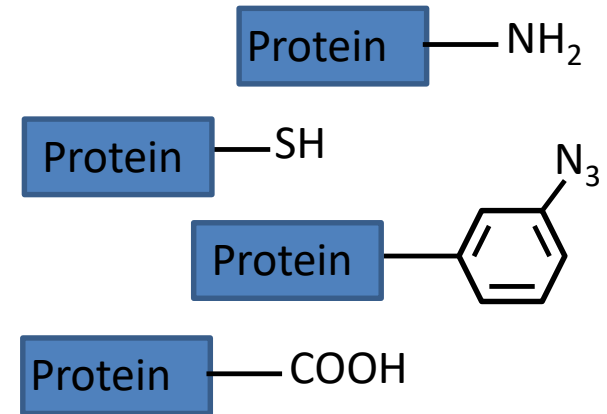
Cross-linked peptide spectra are more complex



Crosslinker chemistry

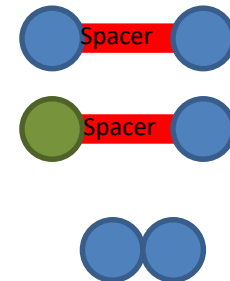
Types

- NH_2 -reactive crosslinker
- Sulfhydryl-reactive crosslinker
- Photoreactive crosslinkers
- COOH -reactive crosslinker



Design:

- Homobifunctional
- Heterobifunctional
- Zero-length



According to:

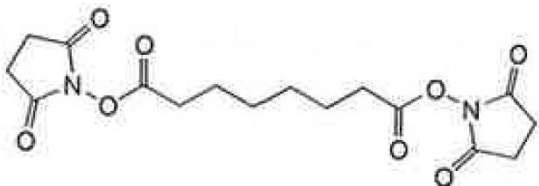
A. Sinz, *Journal of Mass Spectrometry*, 38:1225-1237, 2010

Crosslinkers with different lengths



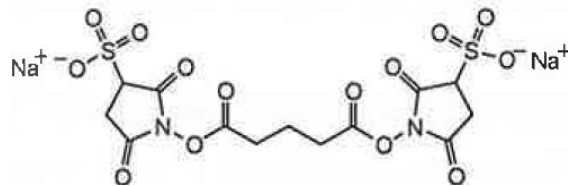
Crosslinker	Chemical name	spacer
DSS	<i>Di</i> (succinimidyl)suberate	11.4 Å
BS2G	<i>Bis</i> (sulfosuccinimidyl)glutarate	7.7 Å
BS2G _{d0/d6}	Di(sulfosuccinimidyl)glutarate	7.7 Å
EDC	1-Ethyl-3-(3-dimethylaminopropyl) carbodiimide hydrochloride	0 Å

DSS



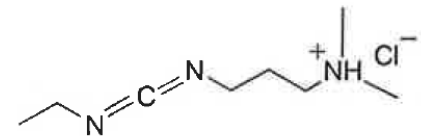
11.4 Å

BS2G



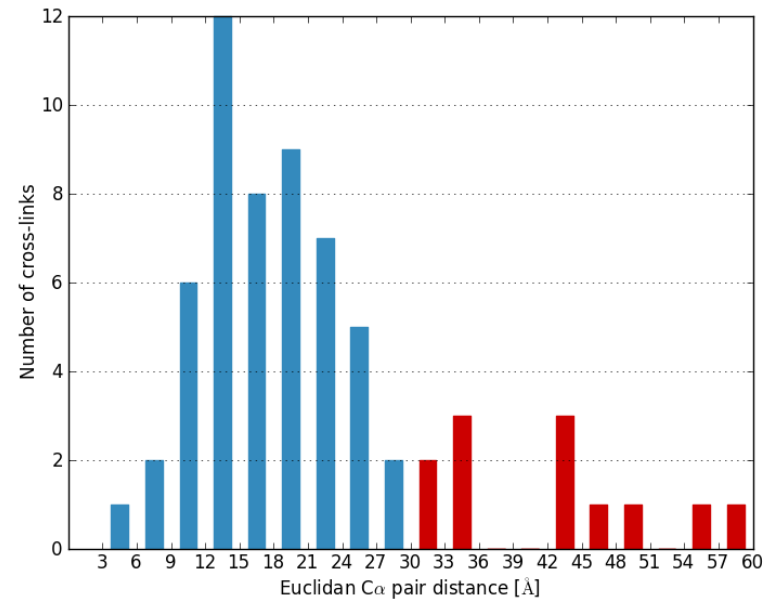
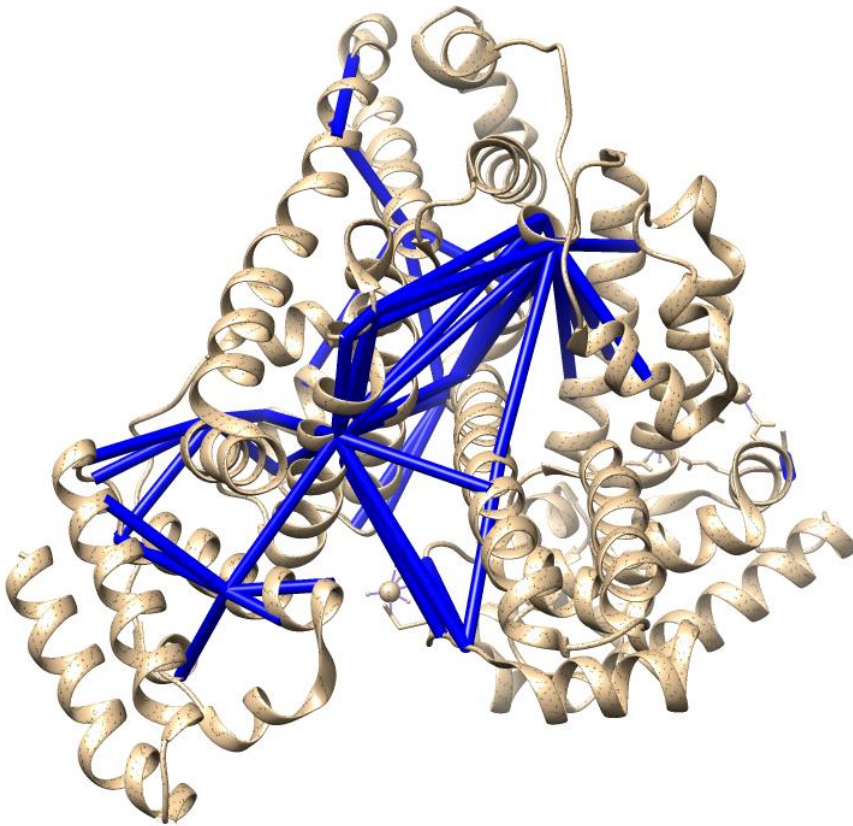
7.7 Å

EDC

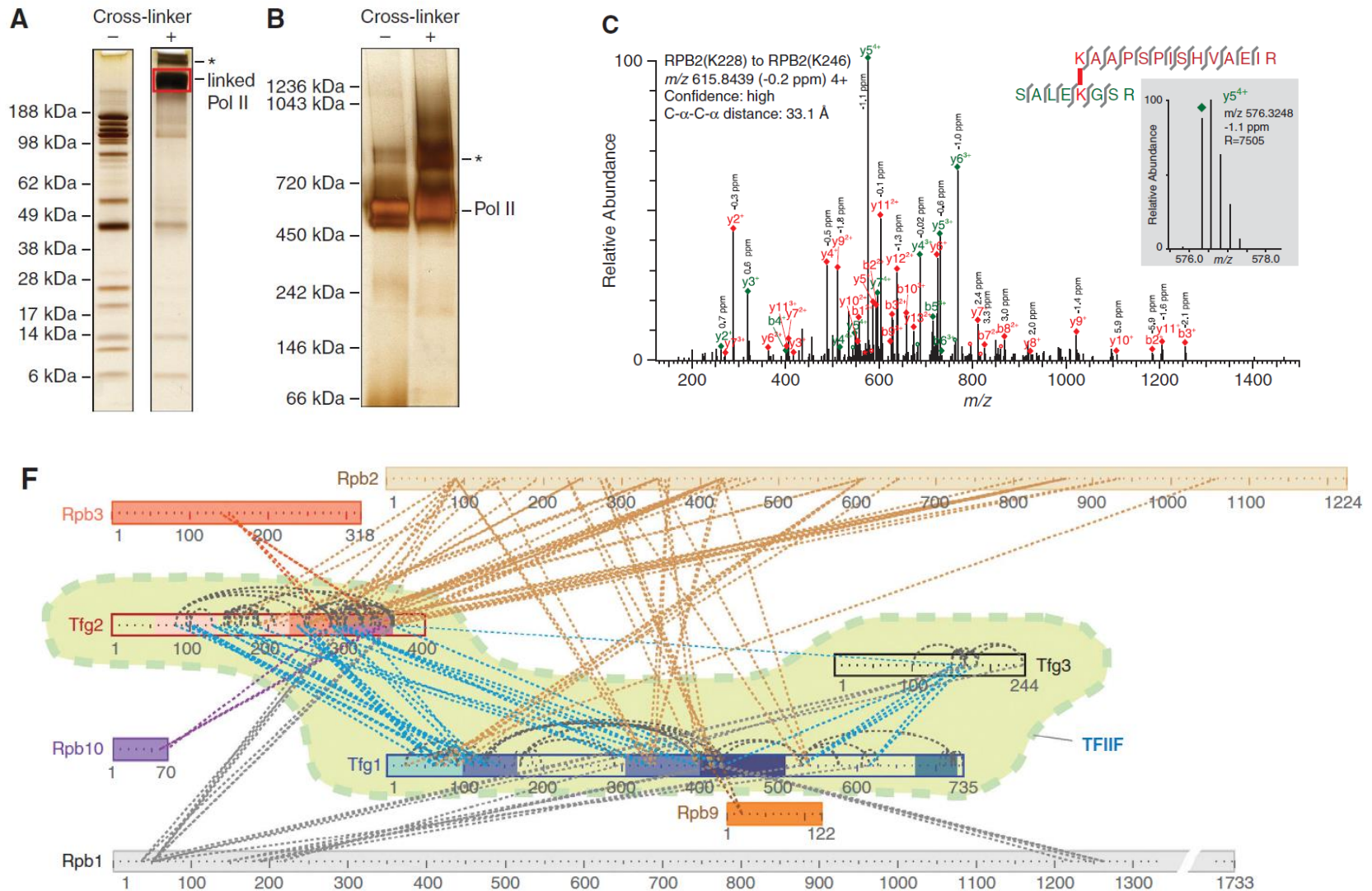


0 Å

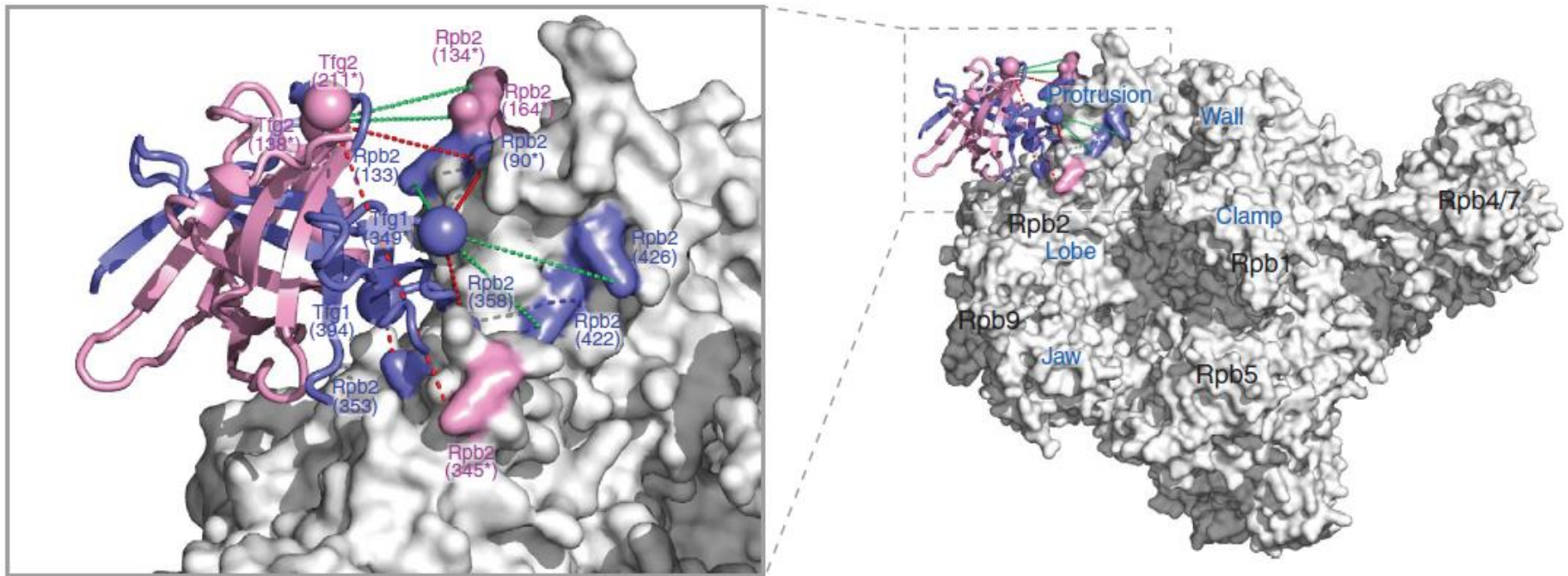
Example results: BSA



Example results: RNA Polymerase II and TFIIF



Binding of TFIIF to RNA Polymerase II

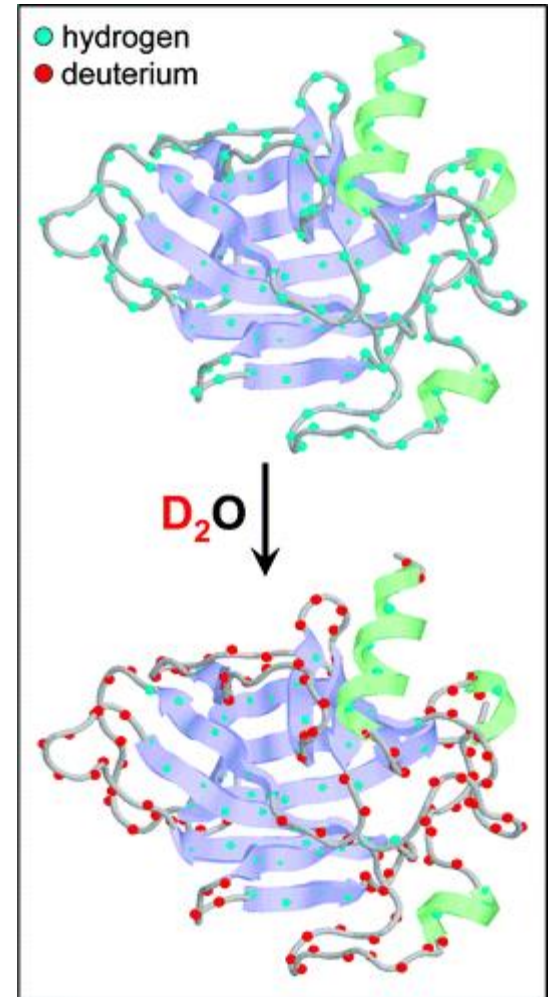


Characterizing structure & interactions with MS

- Identify interaction partners (e.g. Affinity purification MS, BioID, etc.)
- Crosslinking MS
- Hydrogen-deuterium exchange (HDX)
- Native mass spectrometry

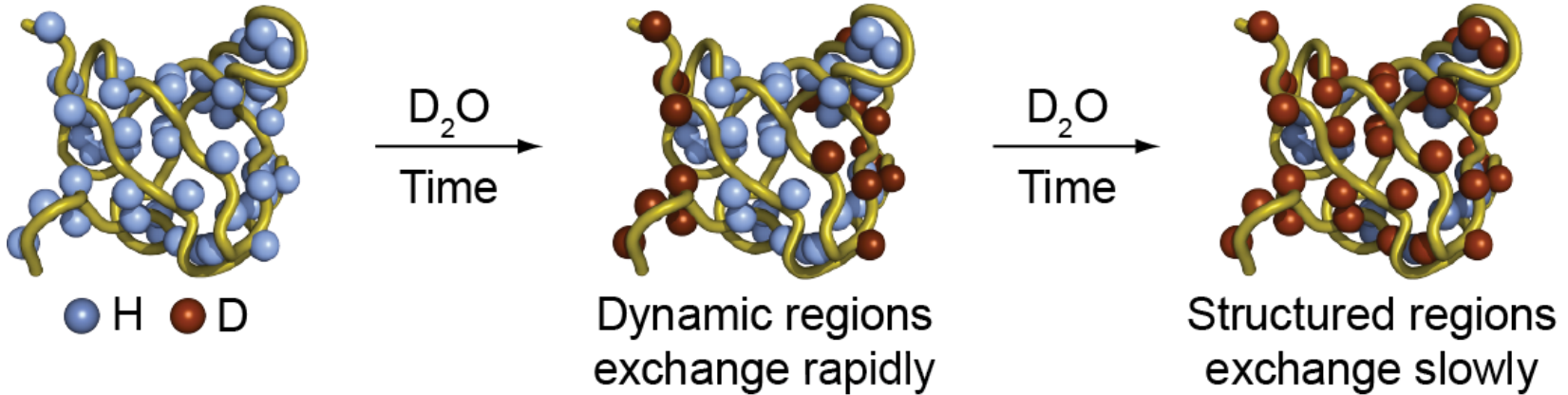
H/D Exchange

- In a solution of D_2O amide-hydrogens can be exchanged against D
- Depends on:
 - Hydrogen bonds
 - Accessibility of H \rightarrow structure!
 - pH of solution
 - temperature

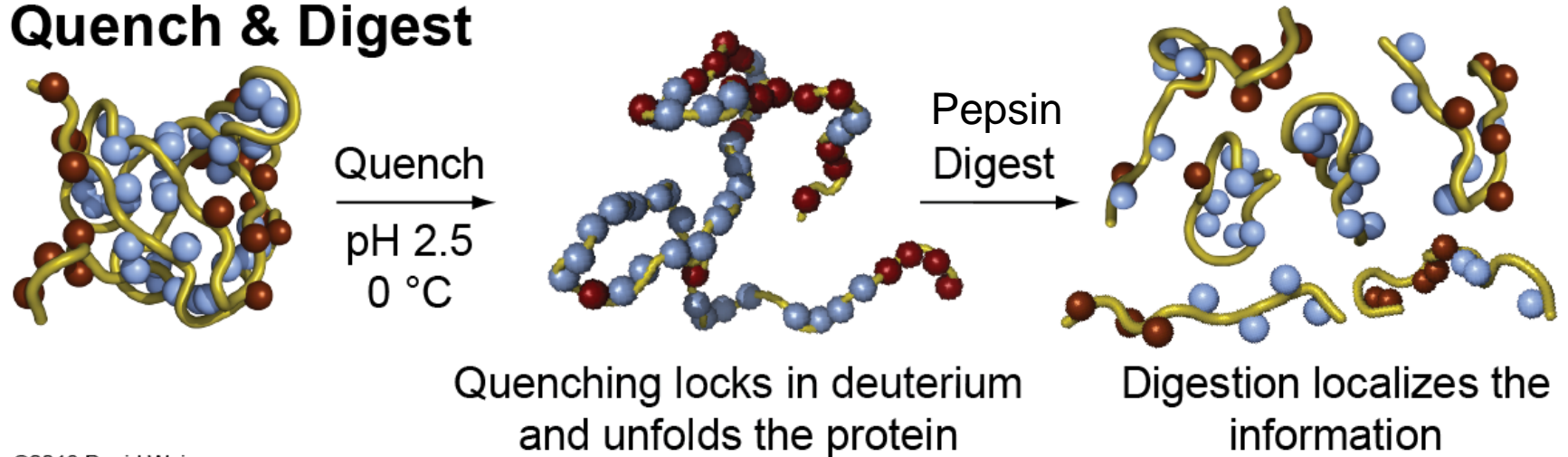


H/D Exchange MS

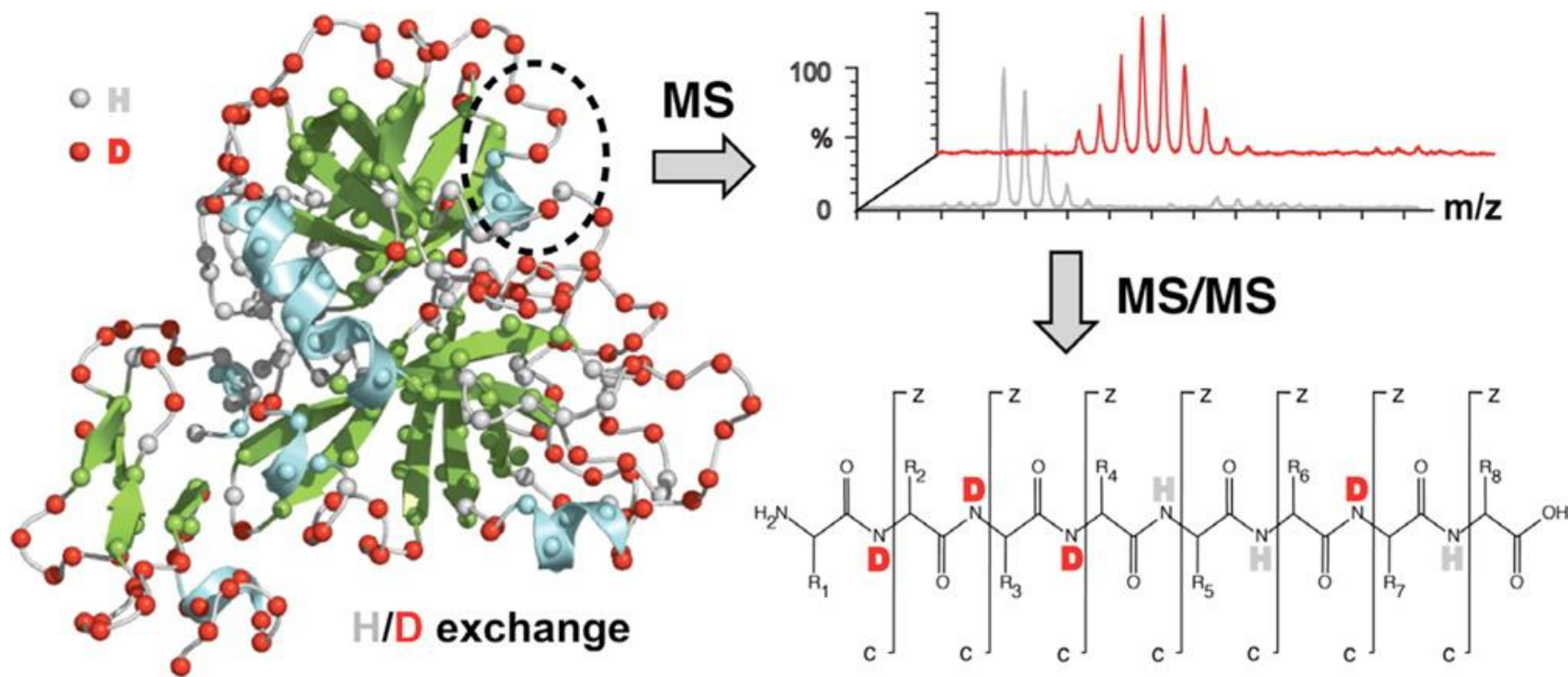
H/D Exchange



Quench & Digest



Amino acid specific information



From: Measuring the Hydrogen/Deuterium Exchange of Proteins at High Spatial Resolution by Mass Spectrometry: Overcoming Gas-Phase Hydrogen/Deuterium Scrambling

Kasper D. Rand, Martin Zehl, and Thomas J. D. Jørgensen

Accounts of Chemical Research **2014** 47 (10), 3018-3027

DOI: 10.1021/ar500194w

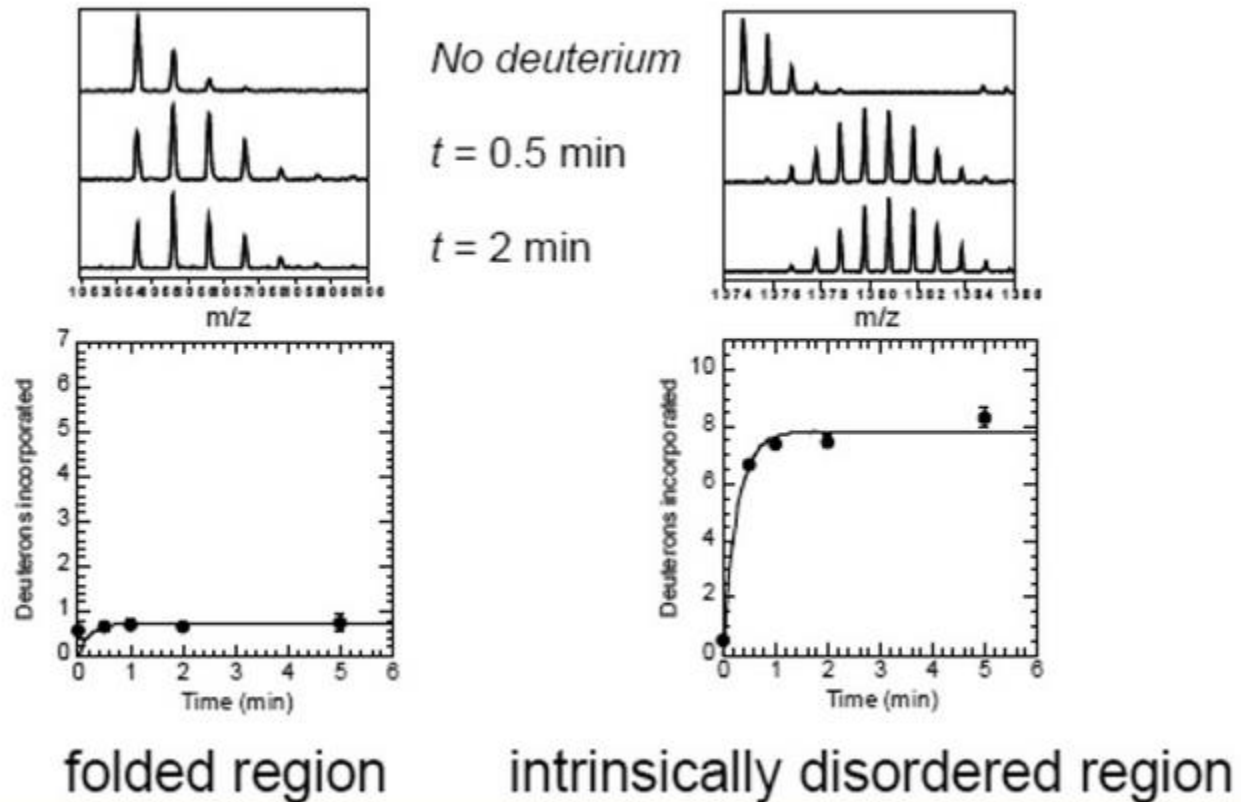
H/D Exchange applications

Provides structural information on:

- Large proteins
- Protein-ligand interaction
- Protein complexes
- Viral particles

Provides information on protein dynamics and conformational state

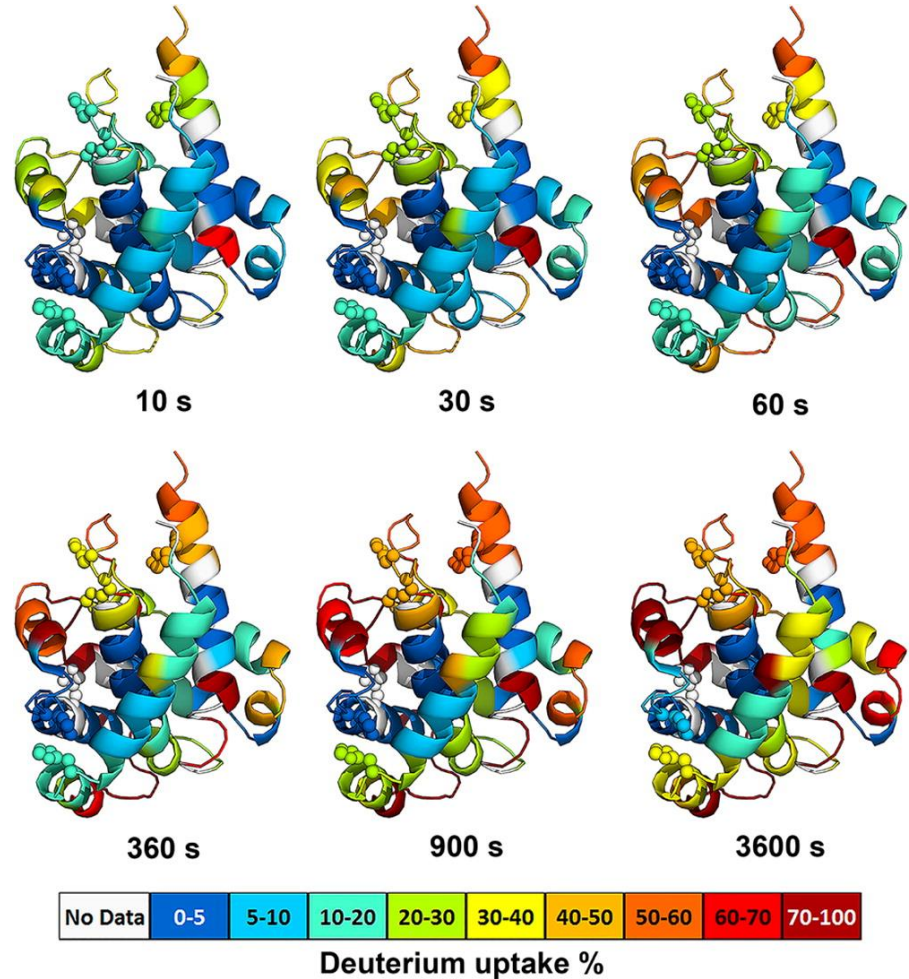
Differentiate folded and disordered regions



H/D exchange time series

Example:

Diphtheria toxin enters cells via endosomal pathway and undergoes a pH dependent conformational change

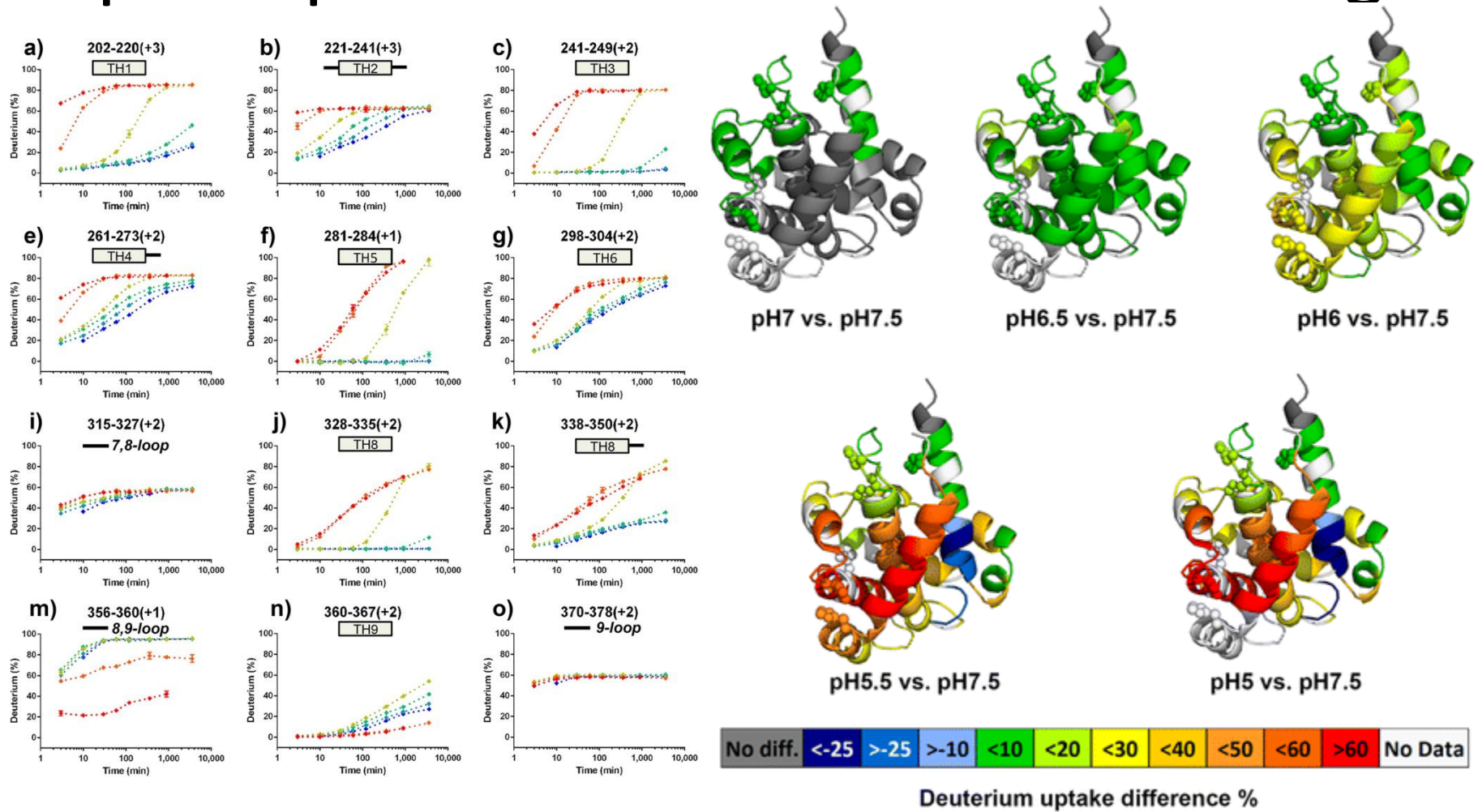


From: Hydrogen–Deuterium Exchange and Mass Spectrometry Reveal the pH-Dependent Conformational Changes of Diphtheria Toxin T Domain

Jing Li, Mykola V. Rodnin, Alexey S. Ladokhin, and Michael L. Gross

Biochemistry 2014 53 (43), 6849-6856

pH dependent conformational change



From: Hydrogen-Deuterium Exchange and Mass Spectrometry Reveal the pH-Dependent Conformational Changes of Diphtheria Toxin T Domain

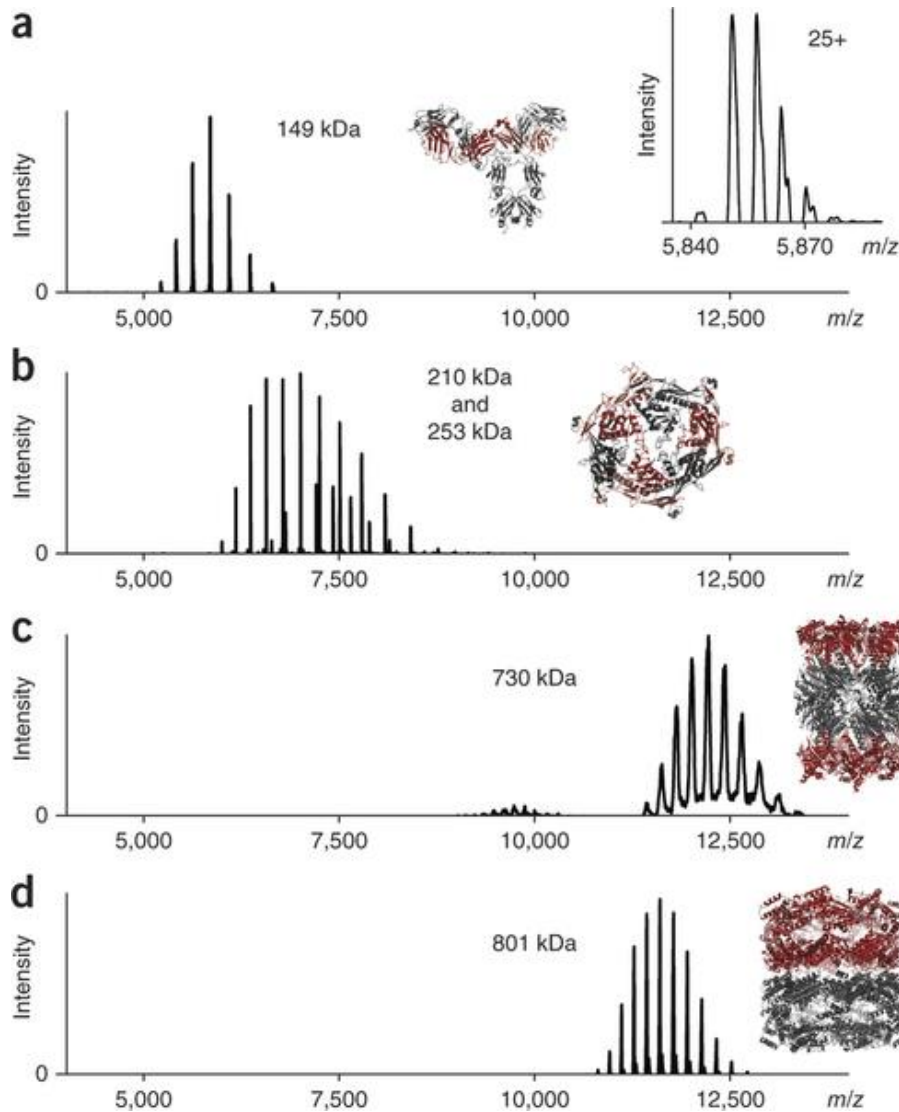
Jing Li, Mykola V. Rodnin, Alexey S. Ladokhin, and Michael L. Gross

Biochemistry 2014 53 (43), 6849-6856

Requirements for HDX

- Pure protein preparations,
- a special LC-MS system that allows digest and separation at low temperatures
- specialised software
- and an experienced application specialist

Back to the start: intact native MS



MS of intact proteins under native conditions (e.g. in ammonium acetate at physiological pH)

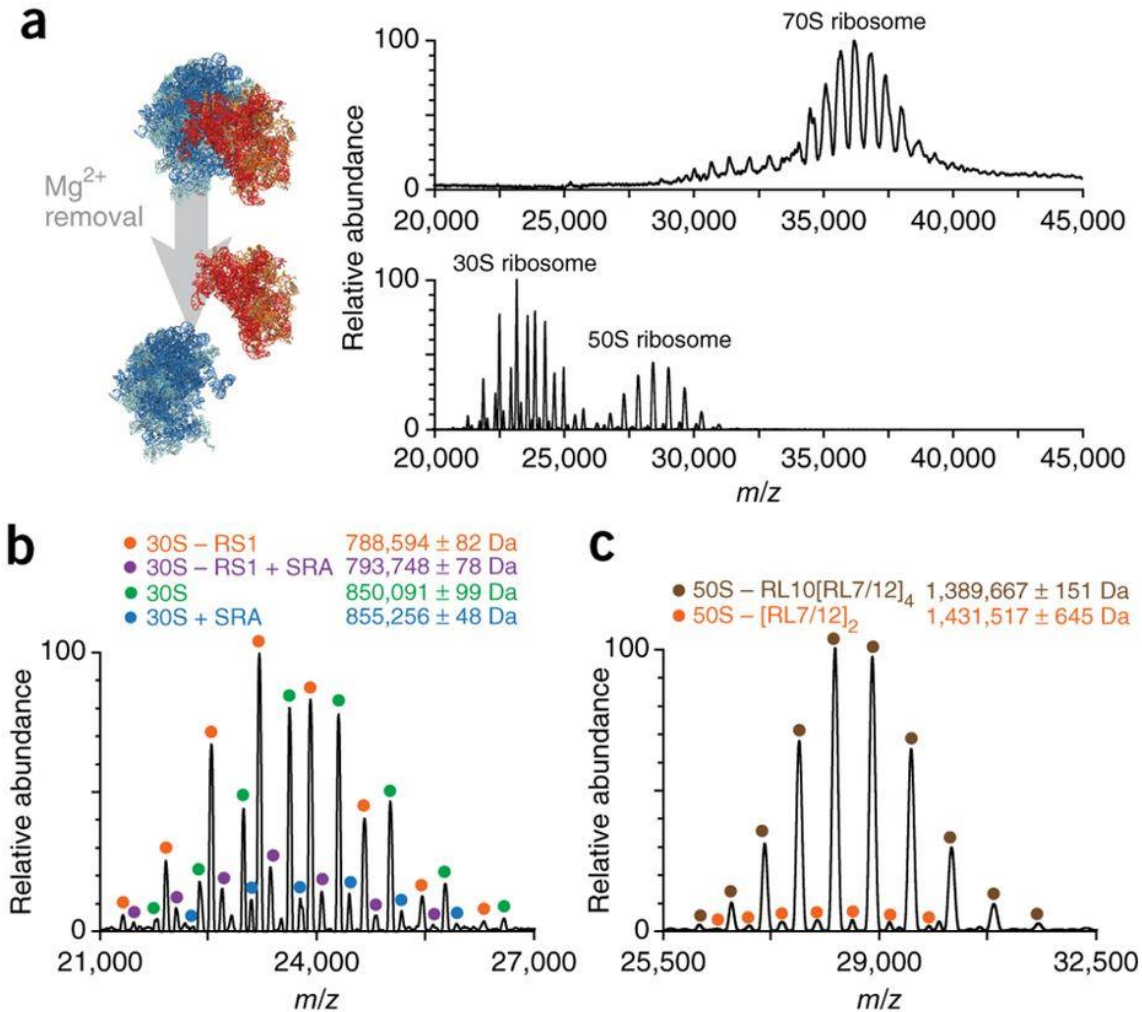
(a–d) Native mass spectra of IgG antibody (a), bacteriophage HK97 capsid pentamers and hexamers (b), yeast 20S proteasome (c) and *E. coli* GroEL (d).

Nat Methods. 2012 Nov;9(11):1084-6. doi: 10.1038/nmeth.2208. Epub 2012 Oct 14.
High-sensitivity Orbitrap mass analysis of intact macromolecular assemblies.
Rose RJ1, Damoc E, Denisov E, Makarov A, Heck AJ.

Native MS

- Buffers that retain native structure (?)
- Samples: min 20 microliters of a 1-5 mg/mL solution
- Mainly manual acquisition

Native MS to the extreme: intact ribosomes



Native MS to the extreme: complexes ejected from native membranes

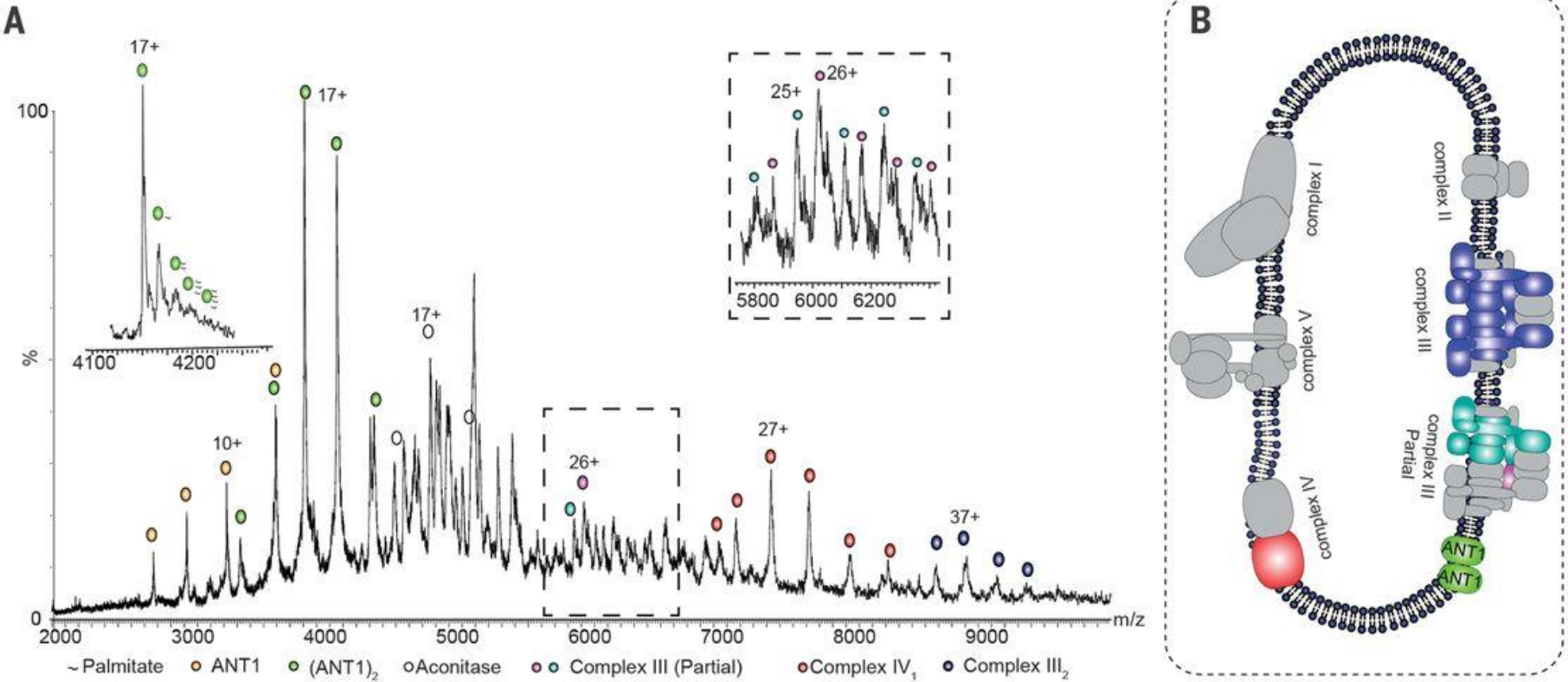
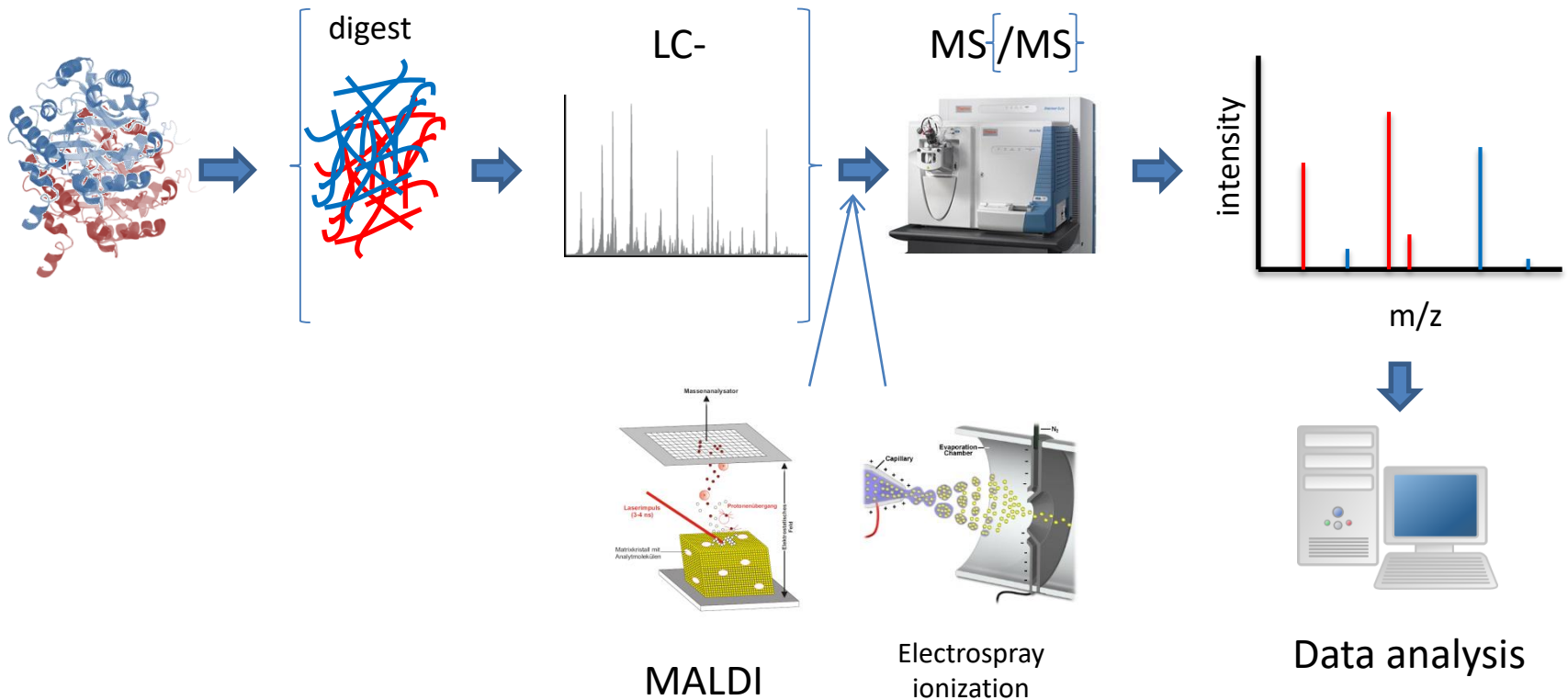


Fig. 3 Intact mitochondria and inner membranes yield complexes I, III, IV, and V, as well as ANT-1 (adenine nucleotide translocase 1) with palmitate transport through the dimer interface.



Summary: Protein mass spectrometry workflow



Summary:

- Basic principles of MS
- Intact vs. Top-down vs. Bottom-up
- MS/MS
- Sequencing & Database searching
- Applications: XLMS, HDX, native MS

Thanks!